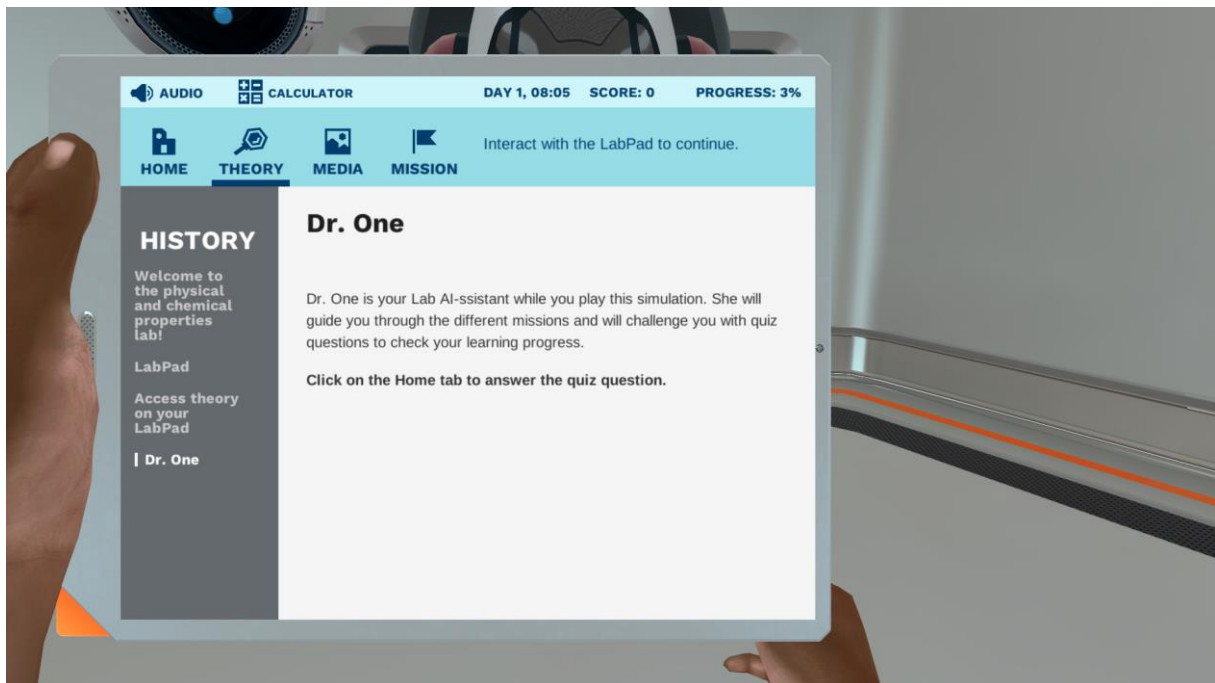
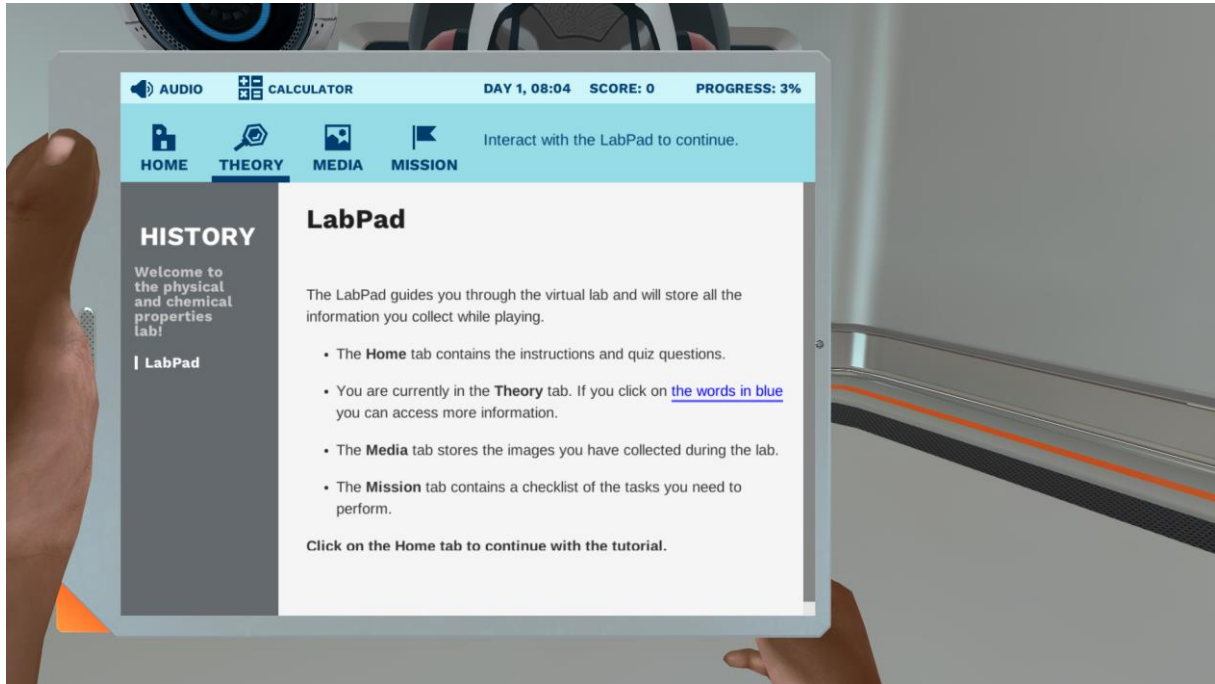
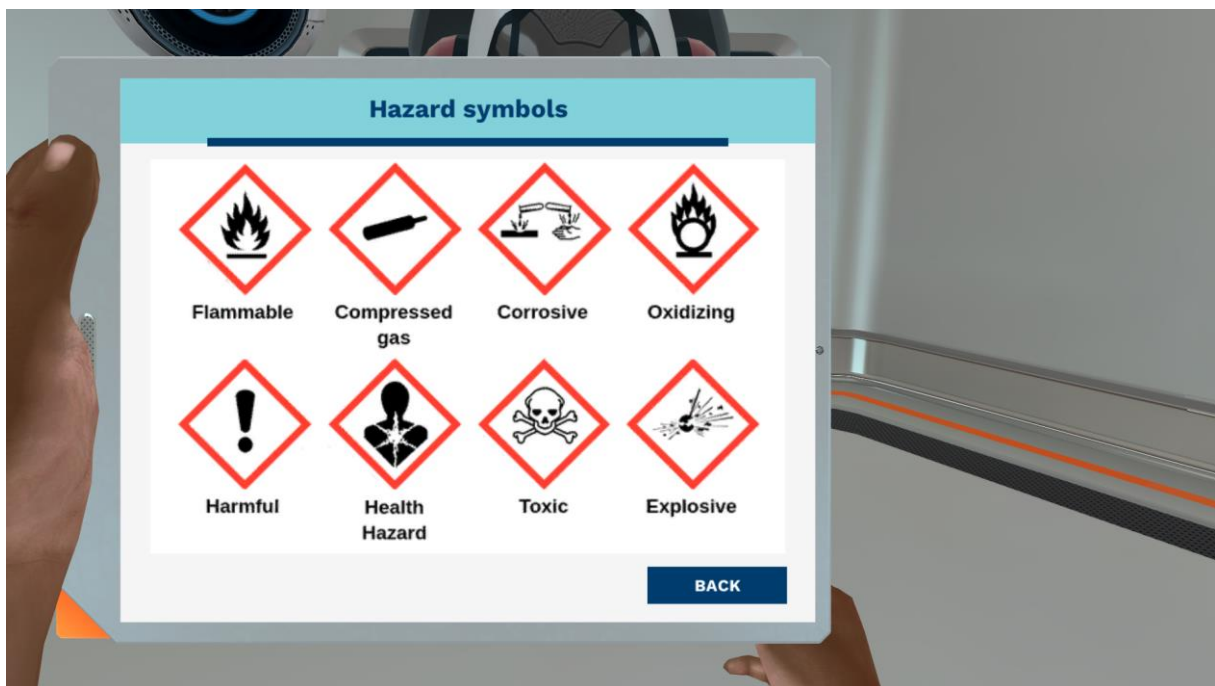
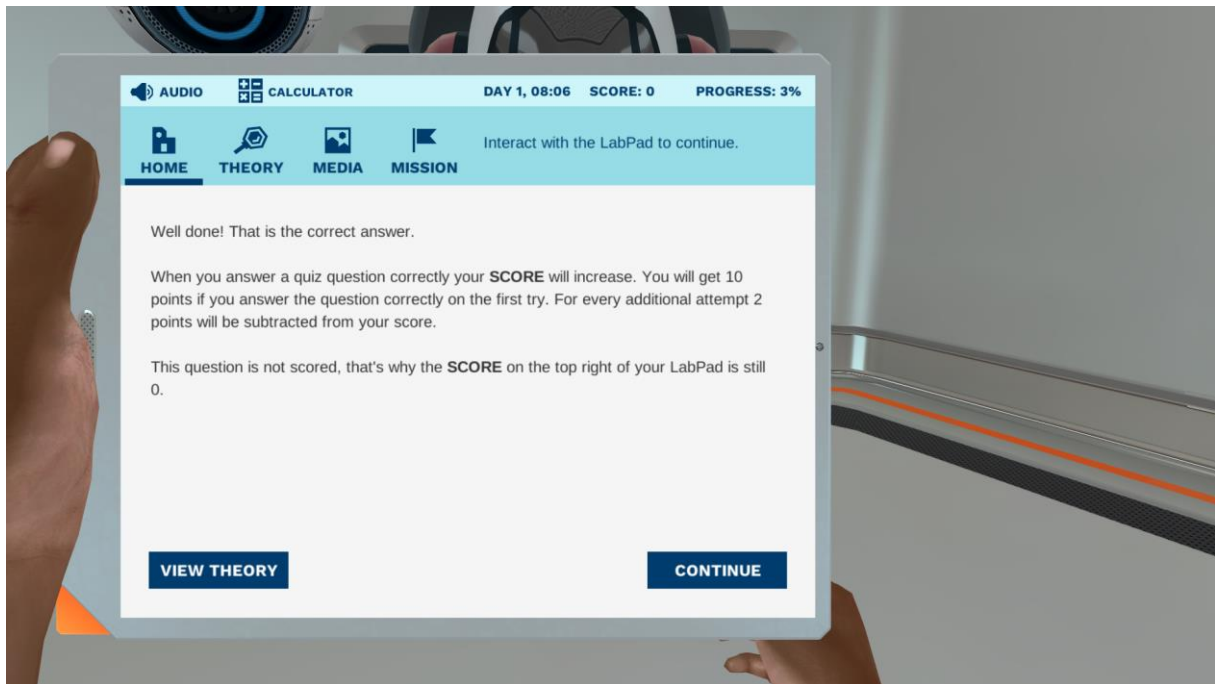
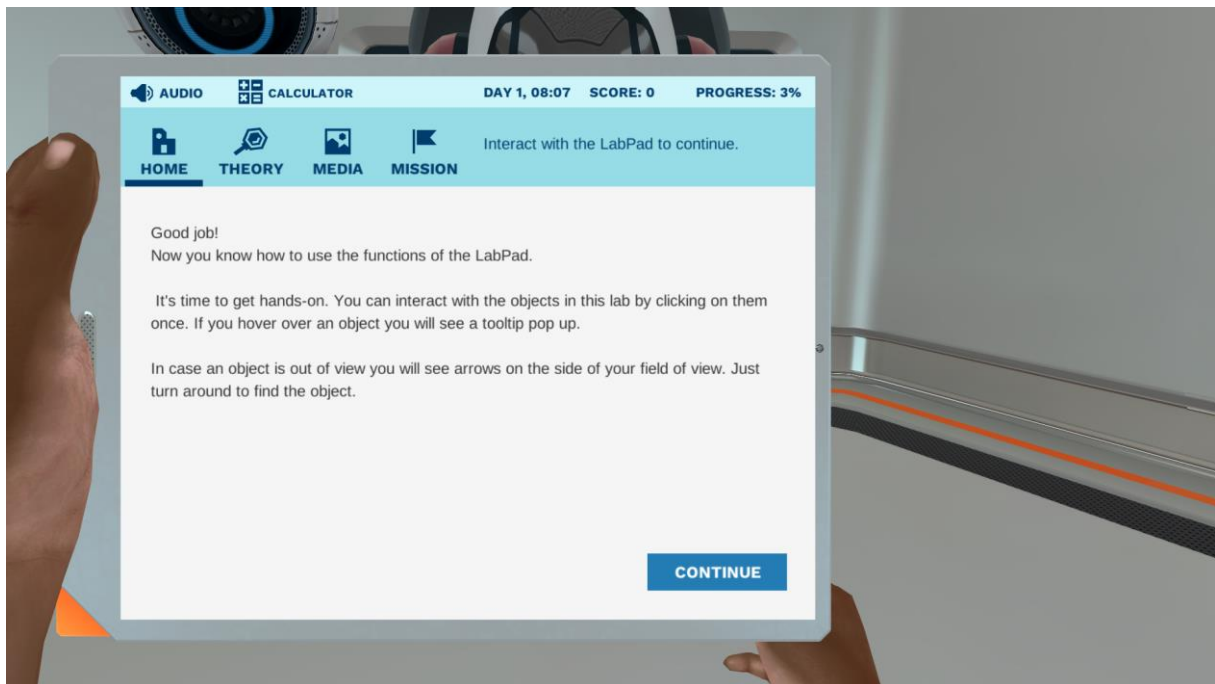
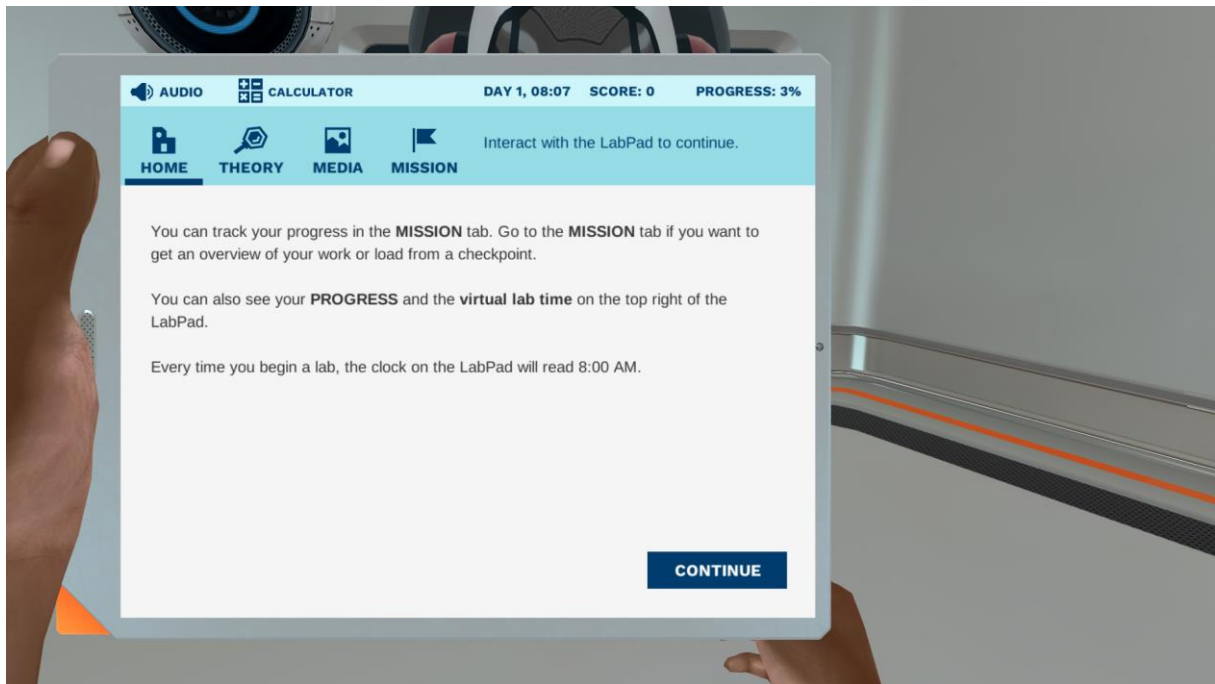


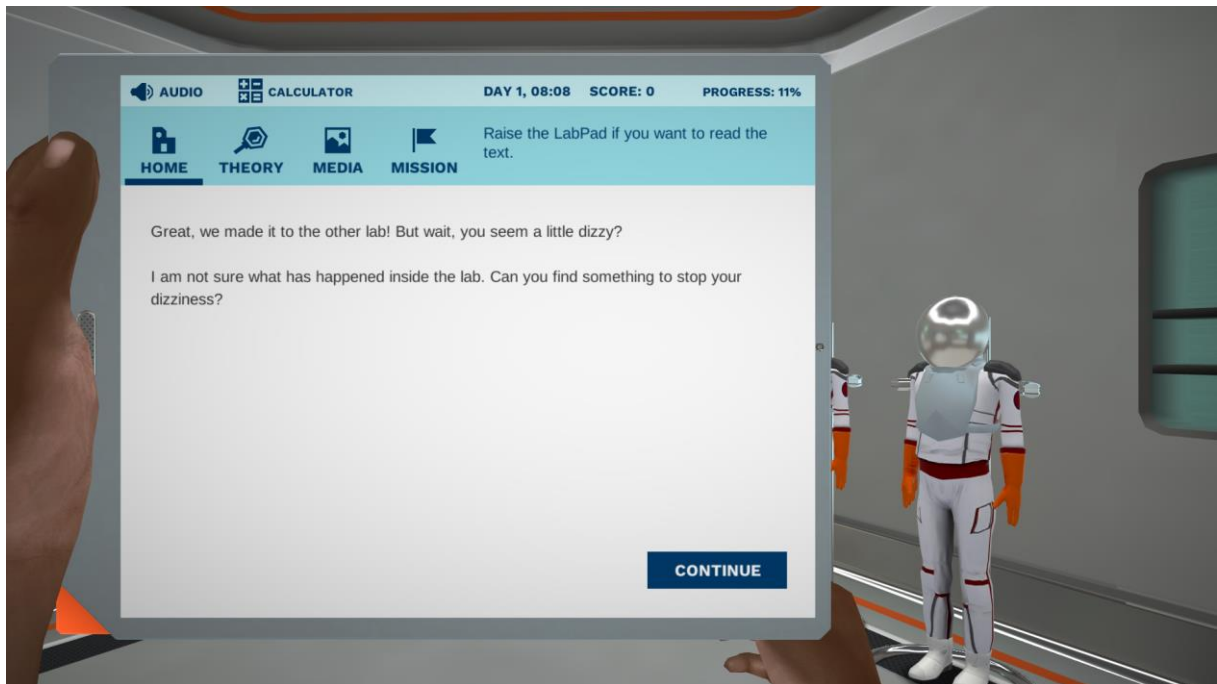
# Sleutel/hulpmiddel/screenshots – labster chemische en fysische eigenschappen

Doorloop eerst de theorie en druk herhaaldelijk op 'CONTINUE'



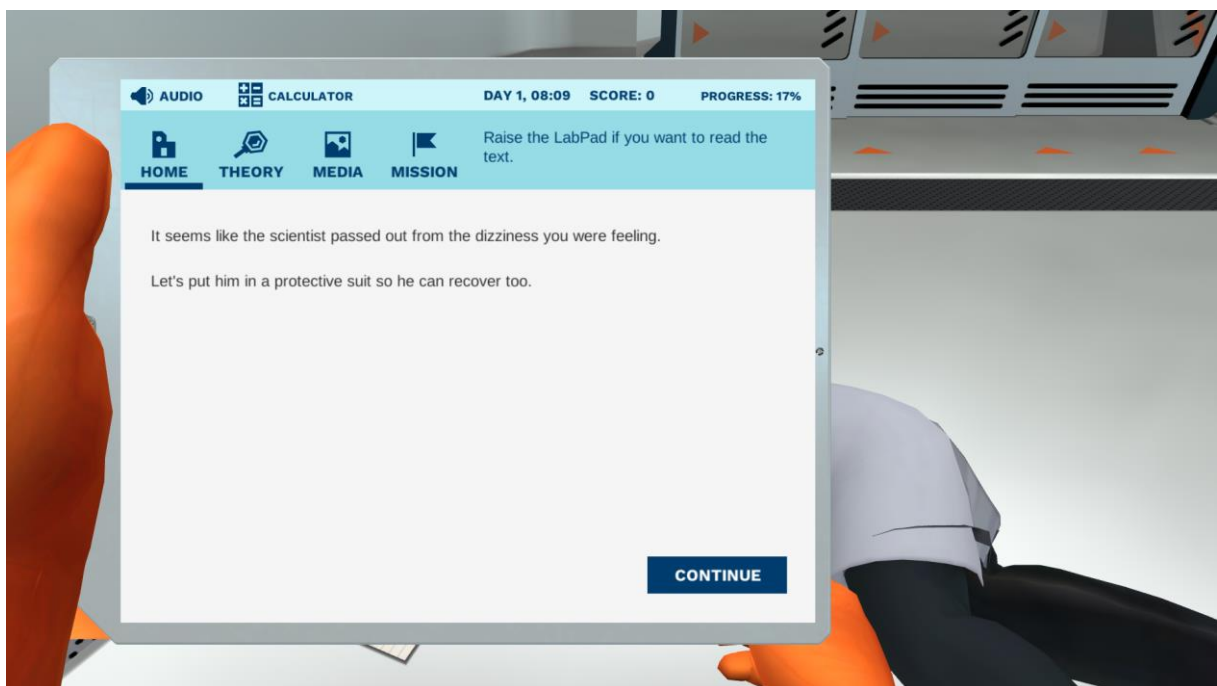




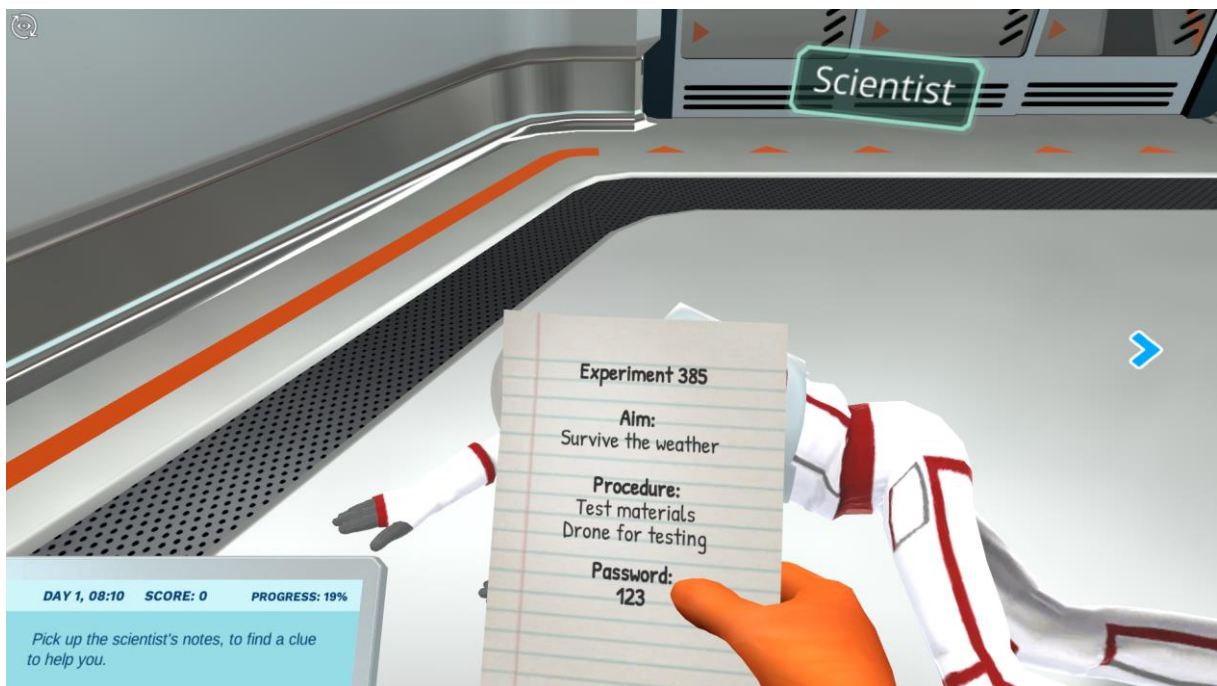
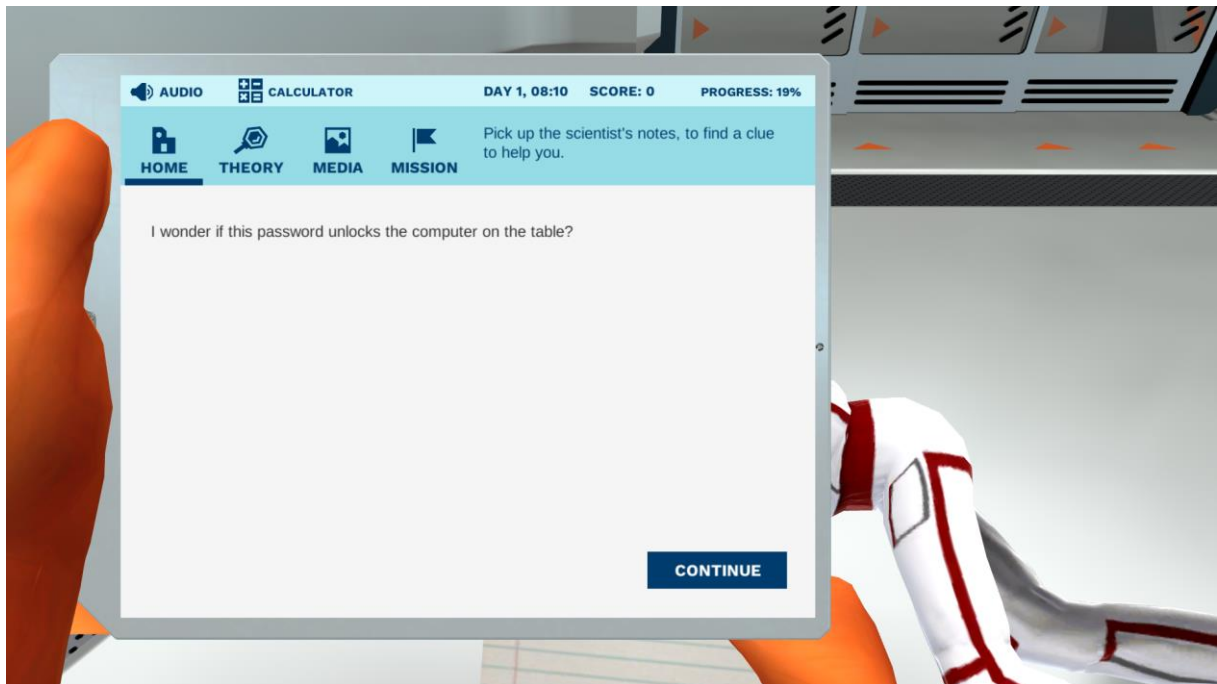


Klik op de ruimtepakken om de duizeligheid op te lossen.

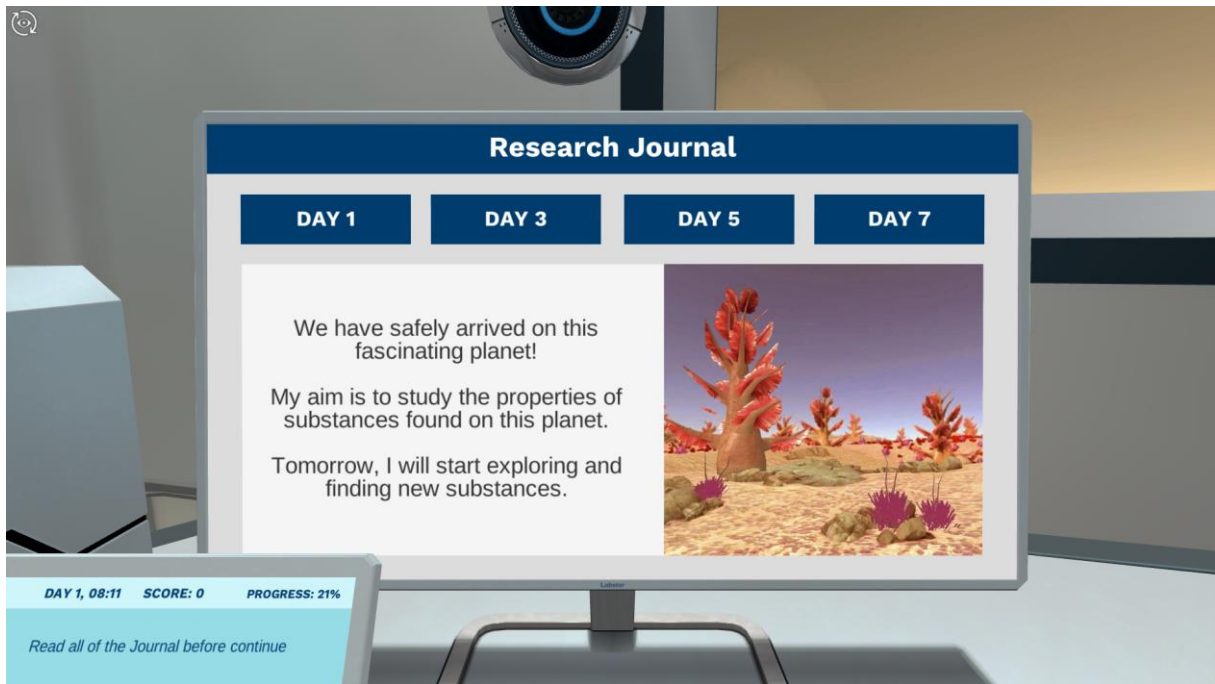
Ga naar de andere ruimte en klik op de wetenschapper.



Neem het bord dat de wetenschapper vast heeft.

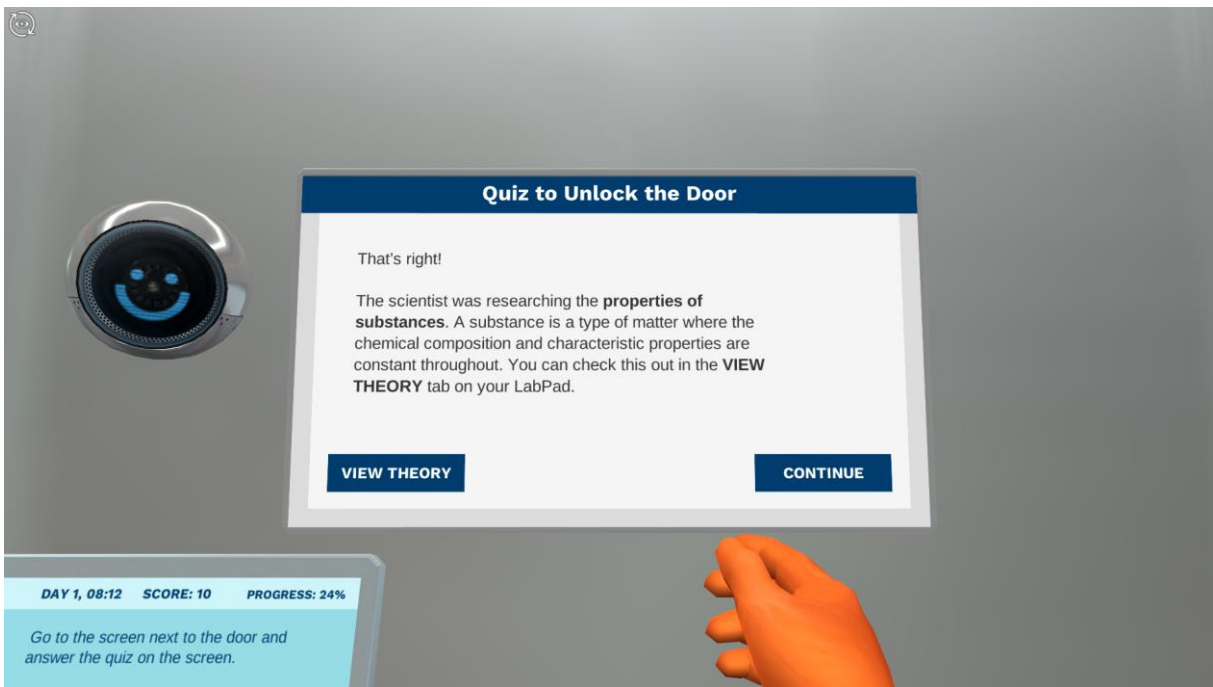


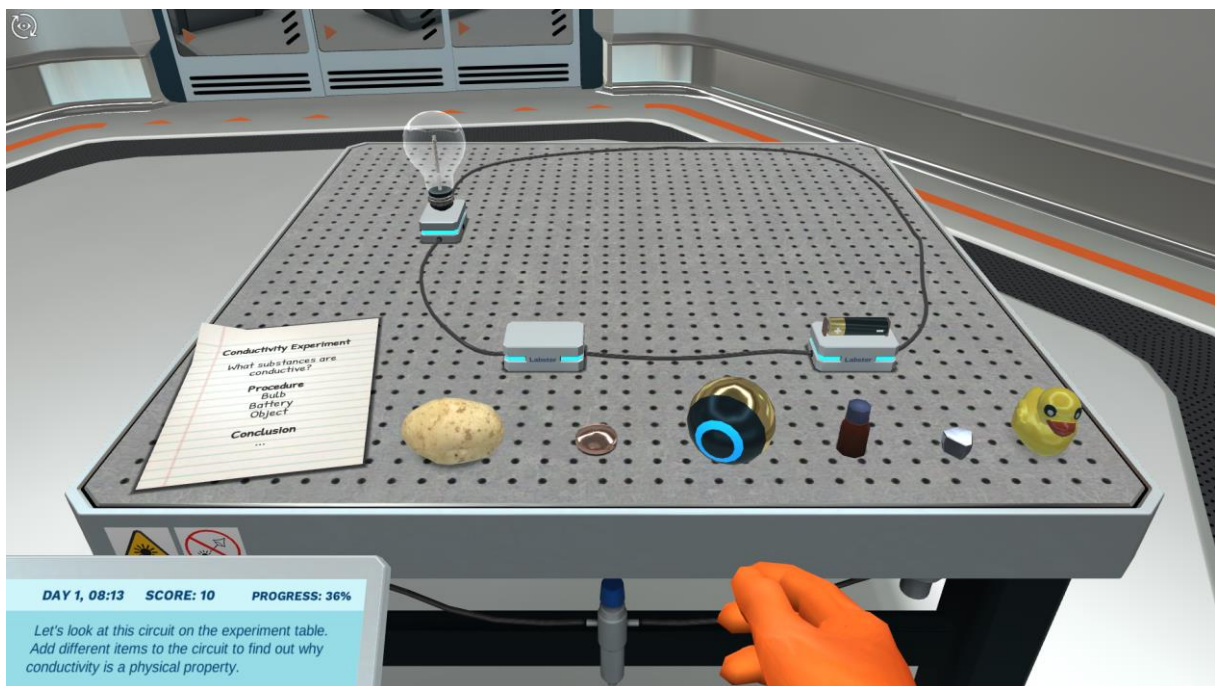
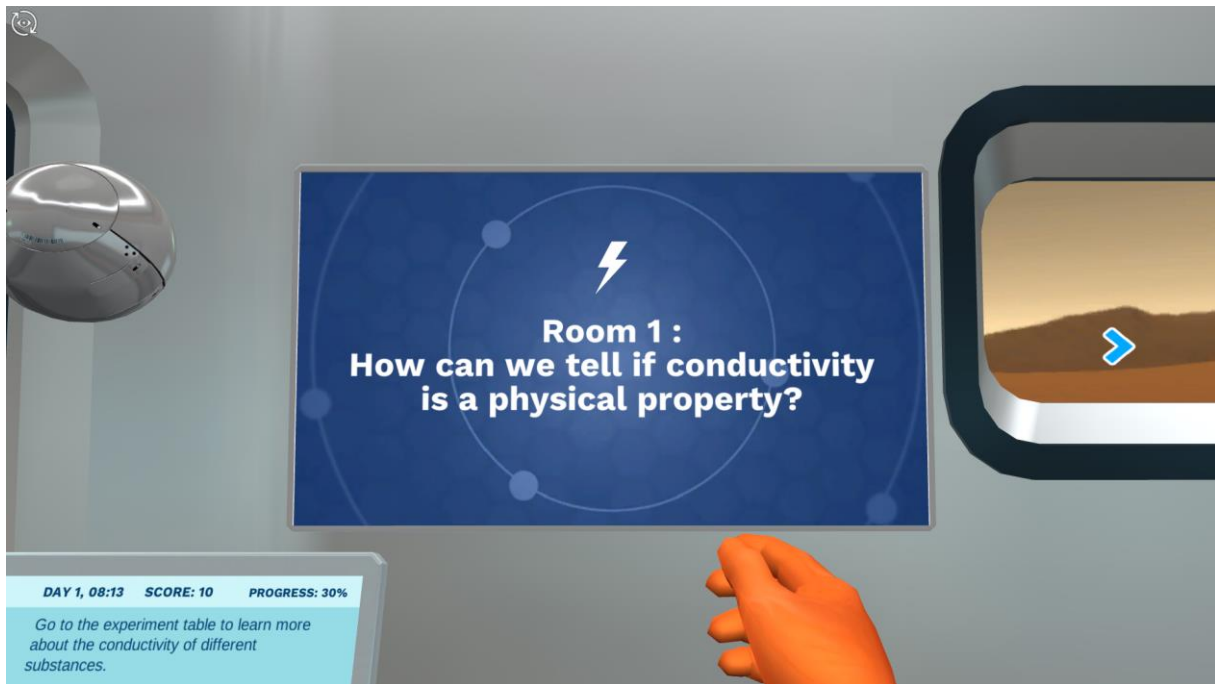
Geef de code (123) in op de computer en druk op enter.



Lees alles in de krant (dag 1,3,5 en 7) om te kunnen antwoorden op de volgende vraag

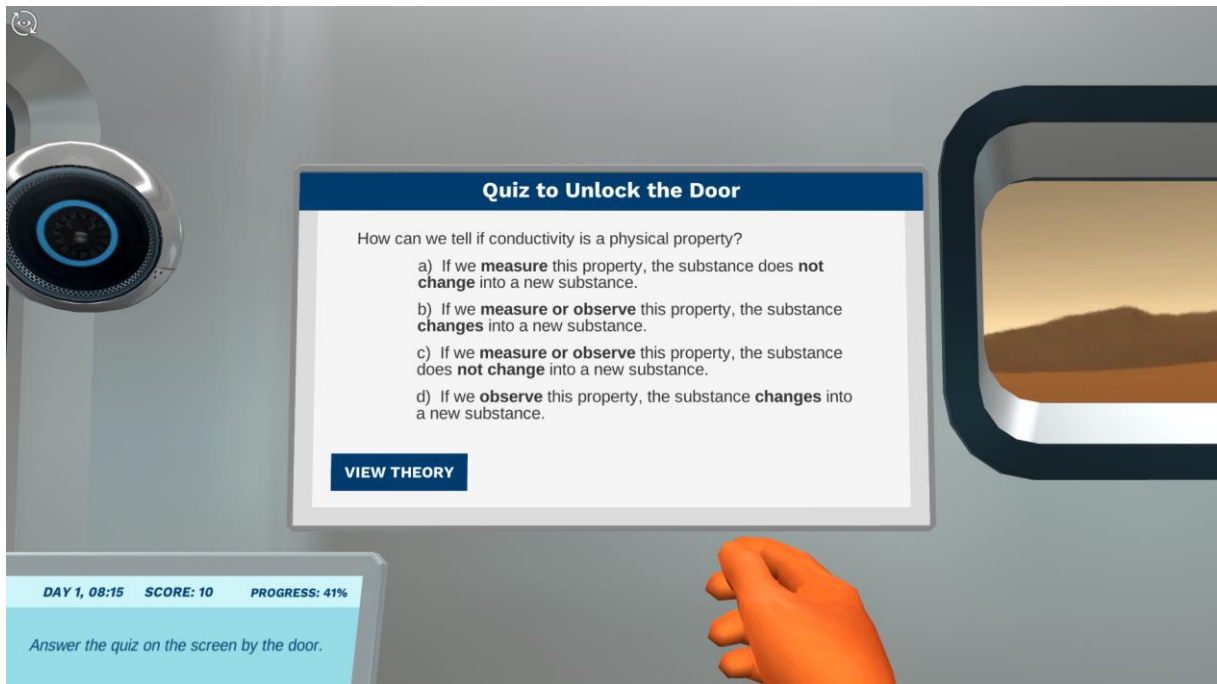




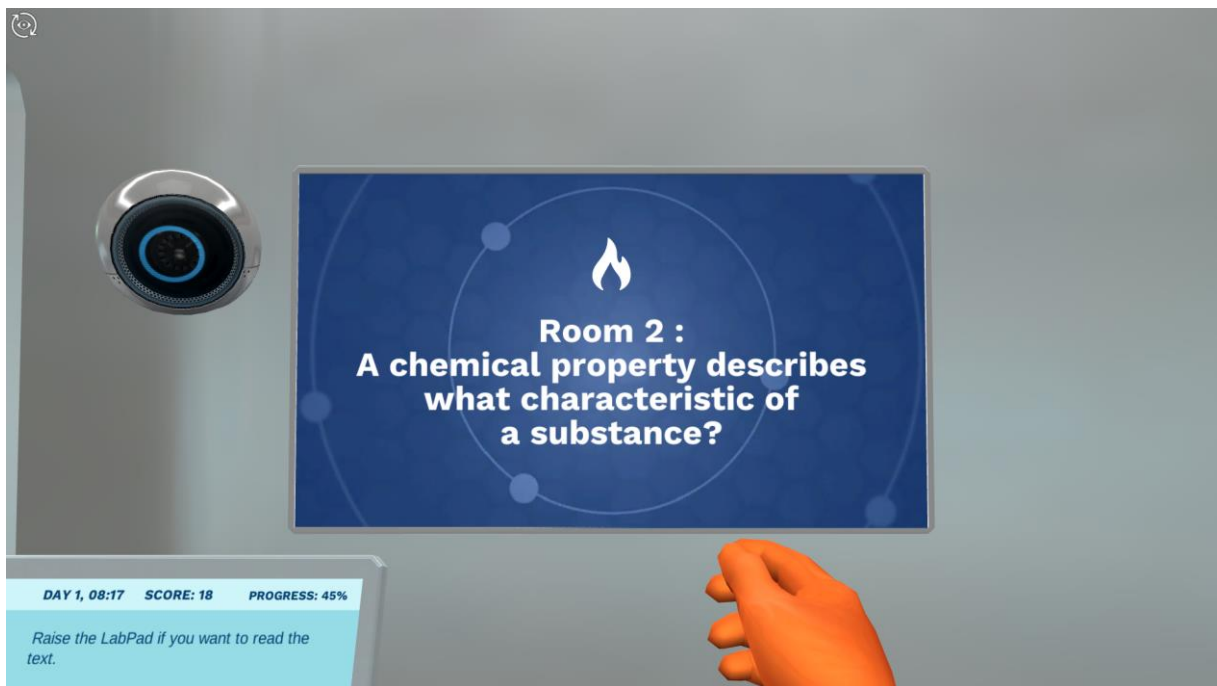


Tik op het voorwerp en plaats het in de kring. Onthoud welk voorwerp de lamp doet branden. Controleer de geleidbaarheid. Tik vervolgens links onder op de blauwe balk.





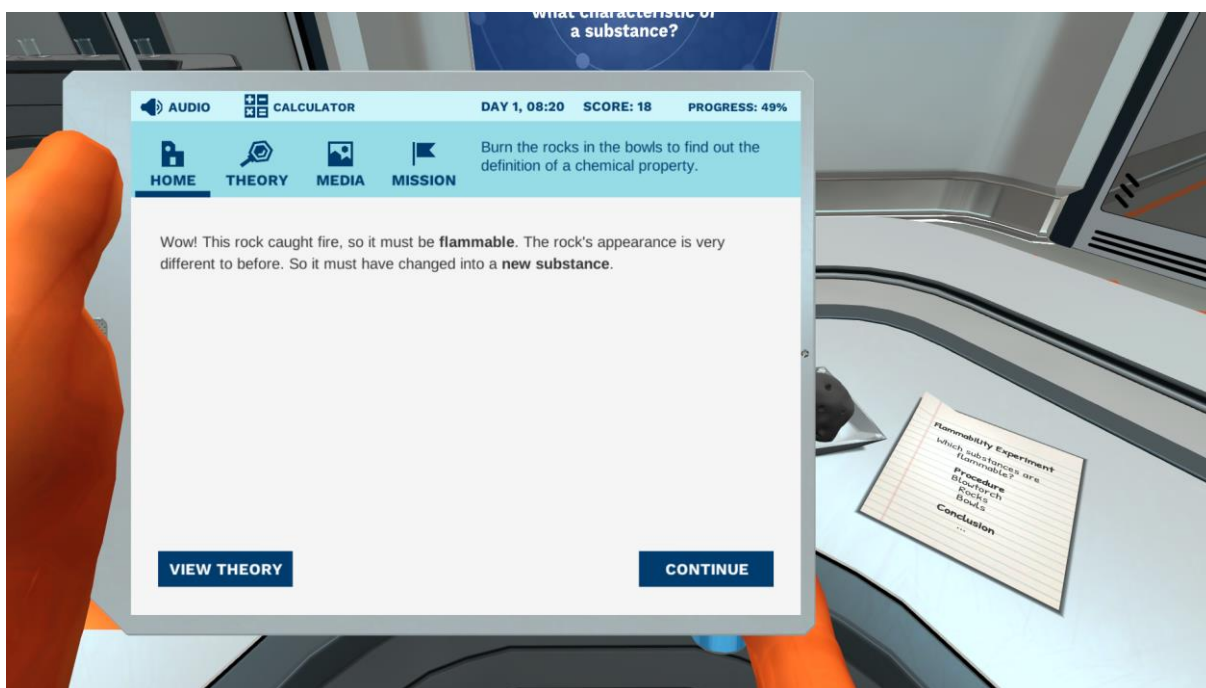
Antwoord c.

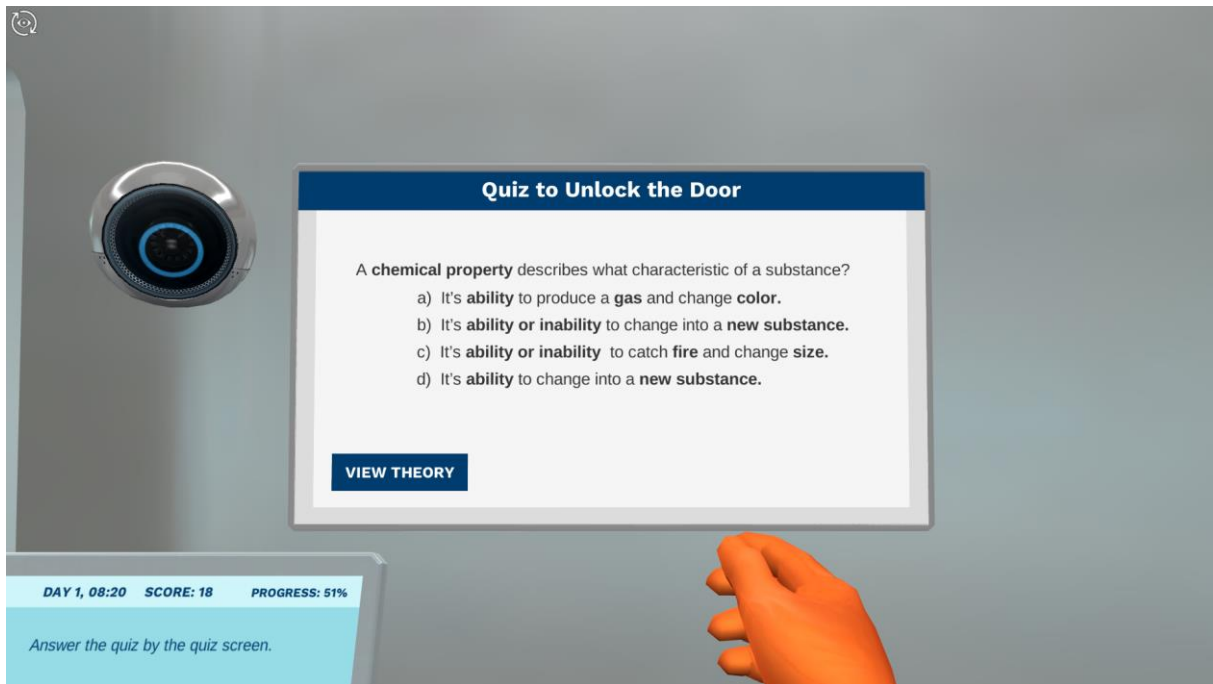




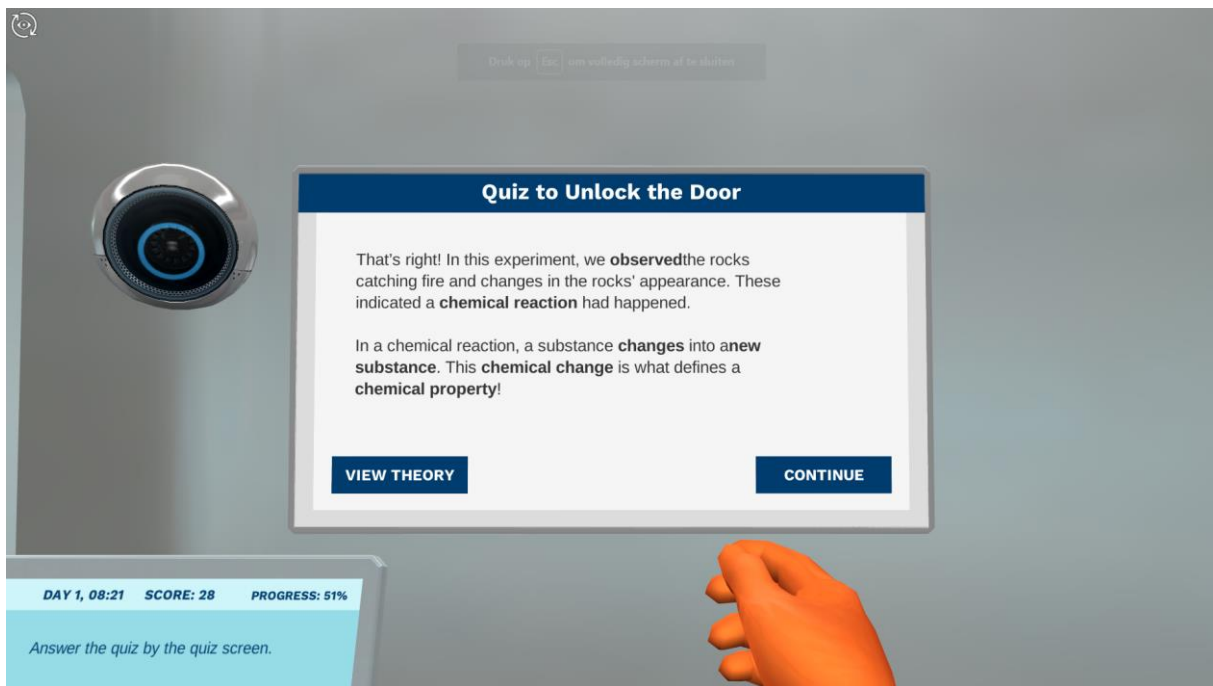
Welke stoffen zijn vlambaar?

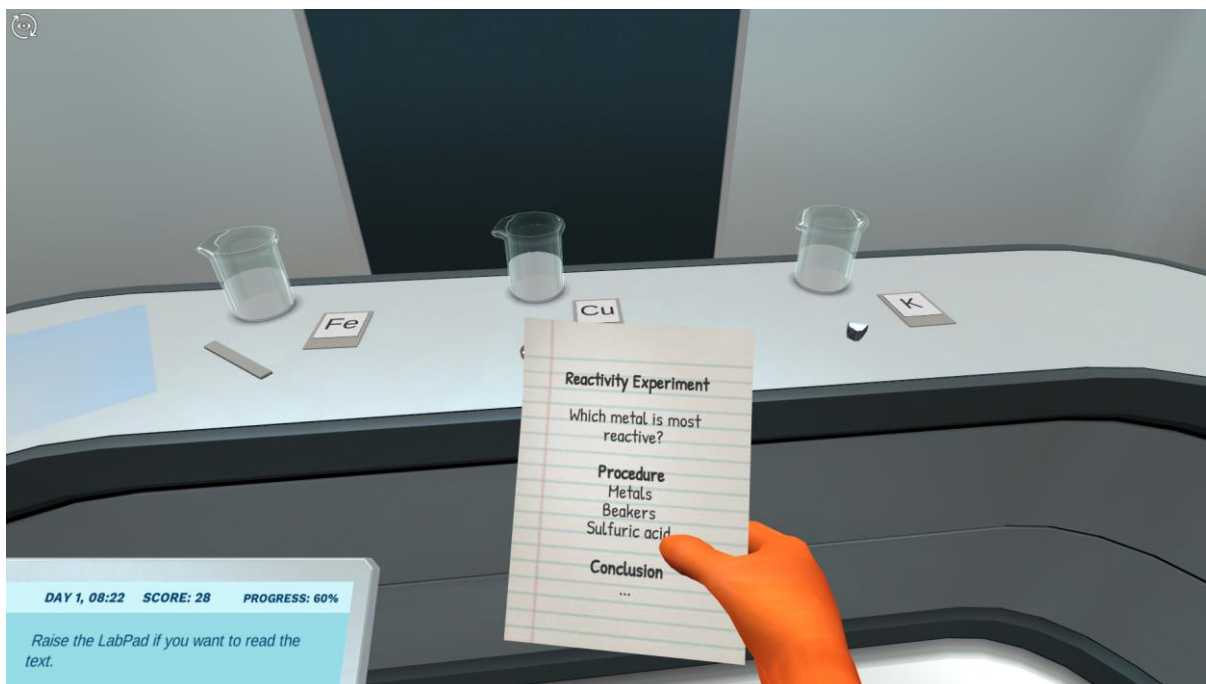
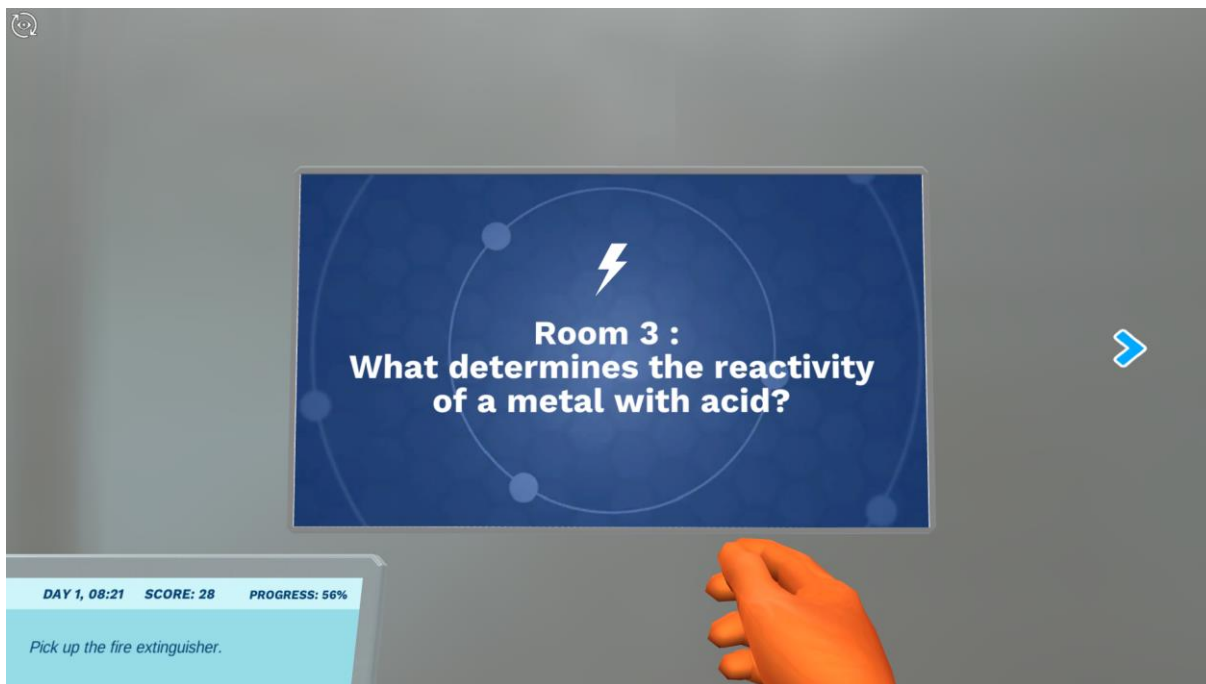
Lees eerst de brief voor je de proef uitvoert.





Antwoord b





Voer de proef uit en rangschik naar reactiviteit

Reactivity of Metals

Reactivity →

Cu	Fe	K
135pm	140pm	220pm

Atomic radius →

DAY 1, 08:23 SCORE: 28 PROGRESS: 64%

Raise the LabPad if you want to read the text.

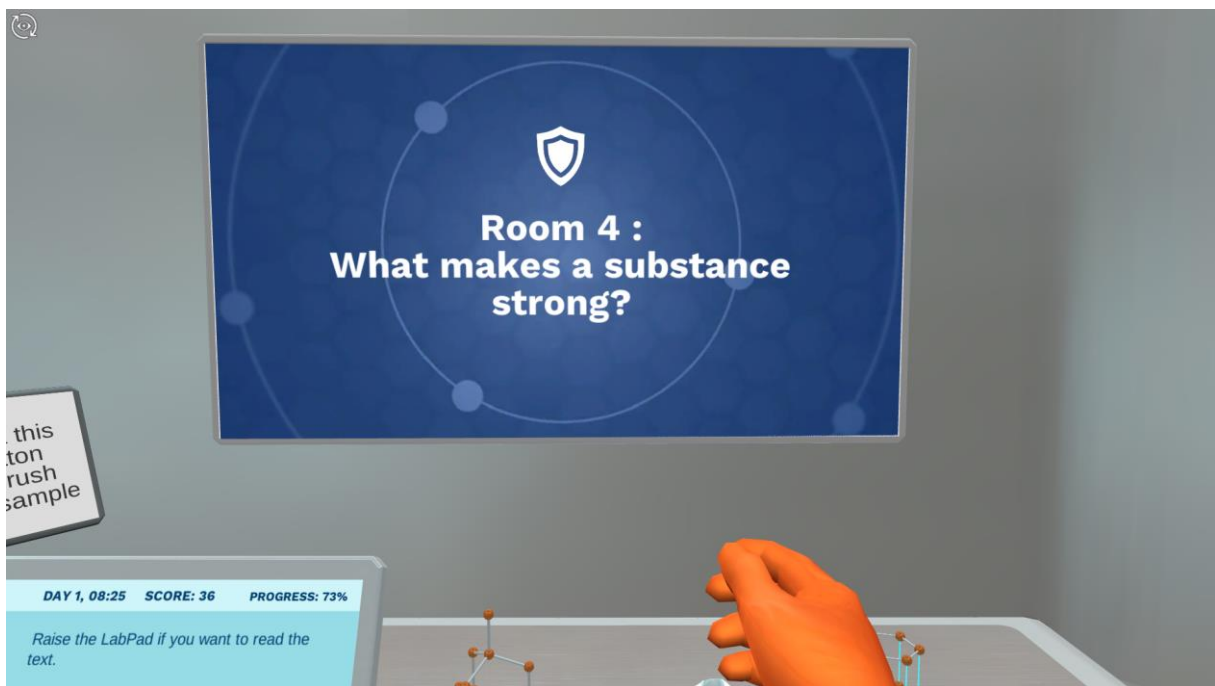
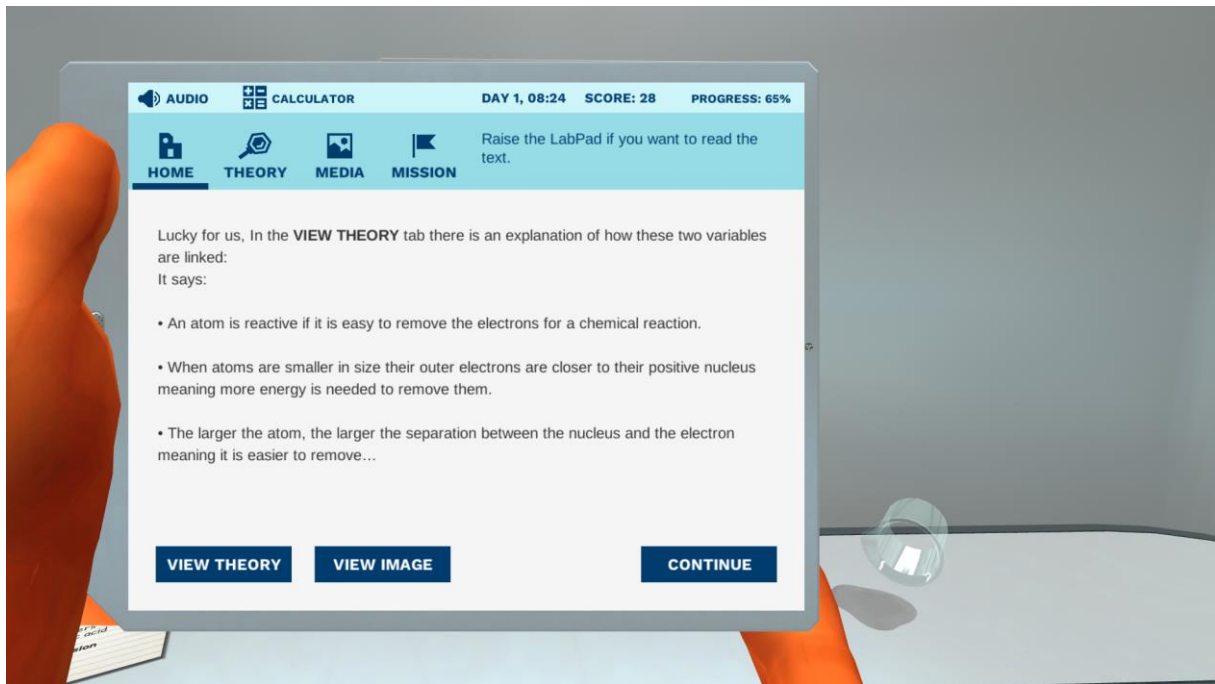
Atom Sizes and Reactivity

Reactivity →

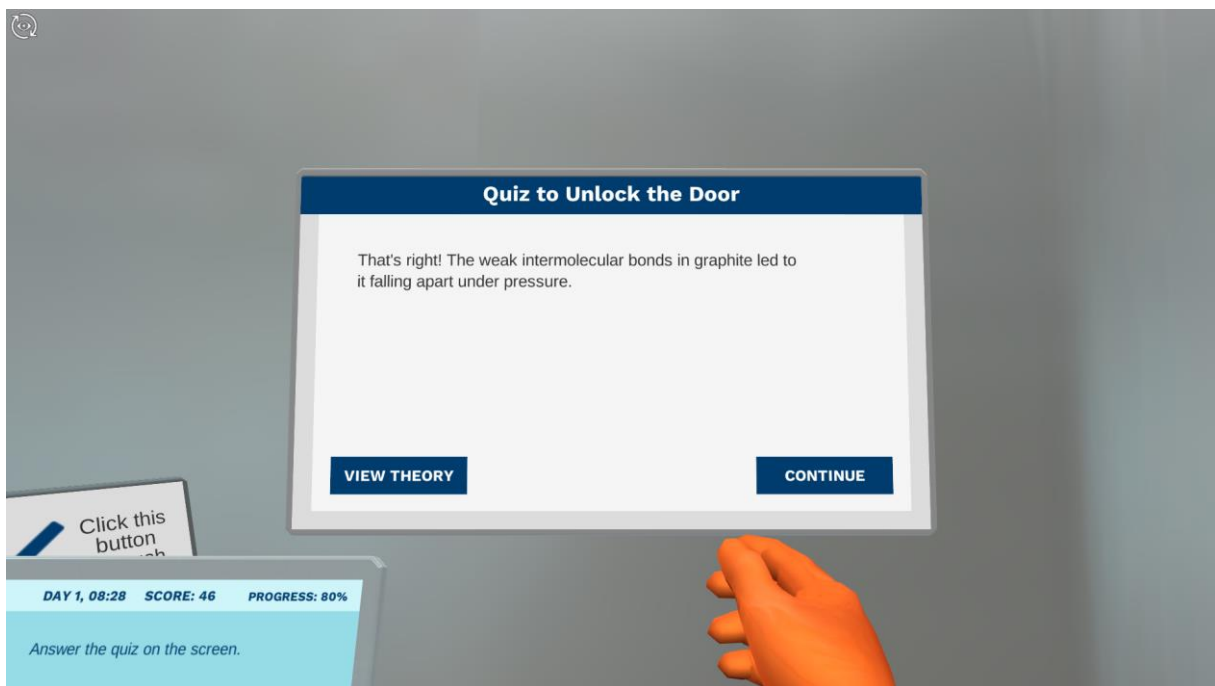
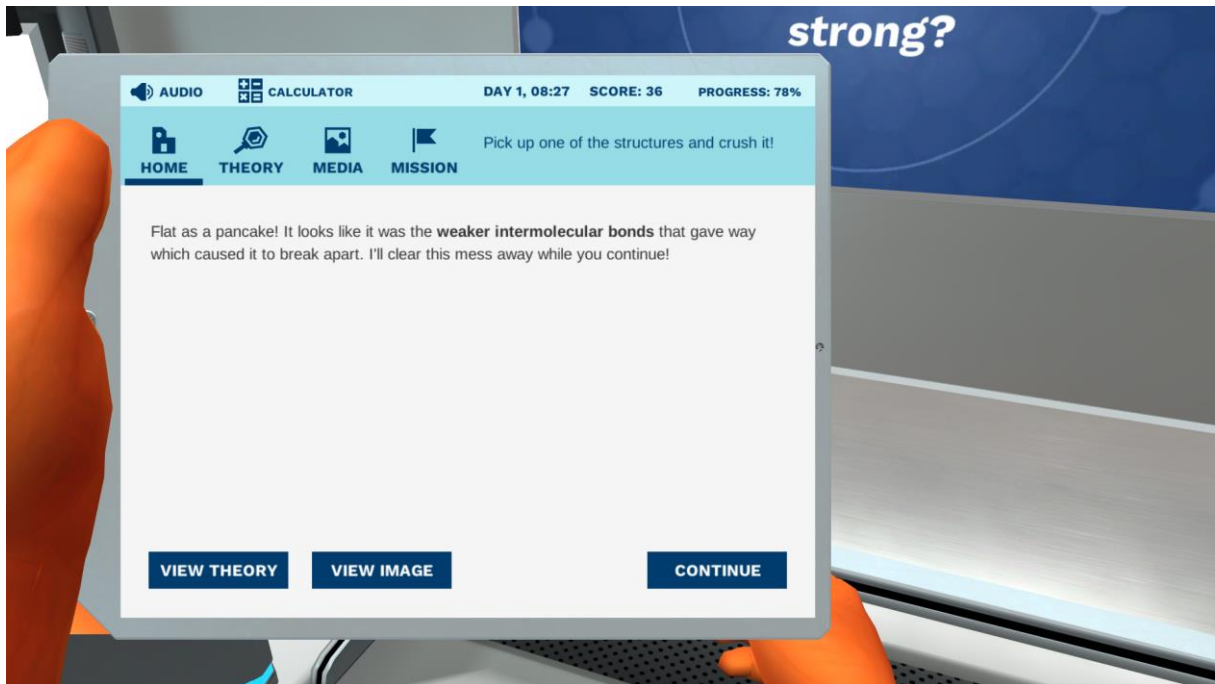
Cu	Fe	K
electron	electron	electron

Atomic radius →

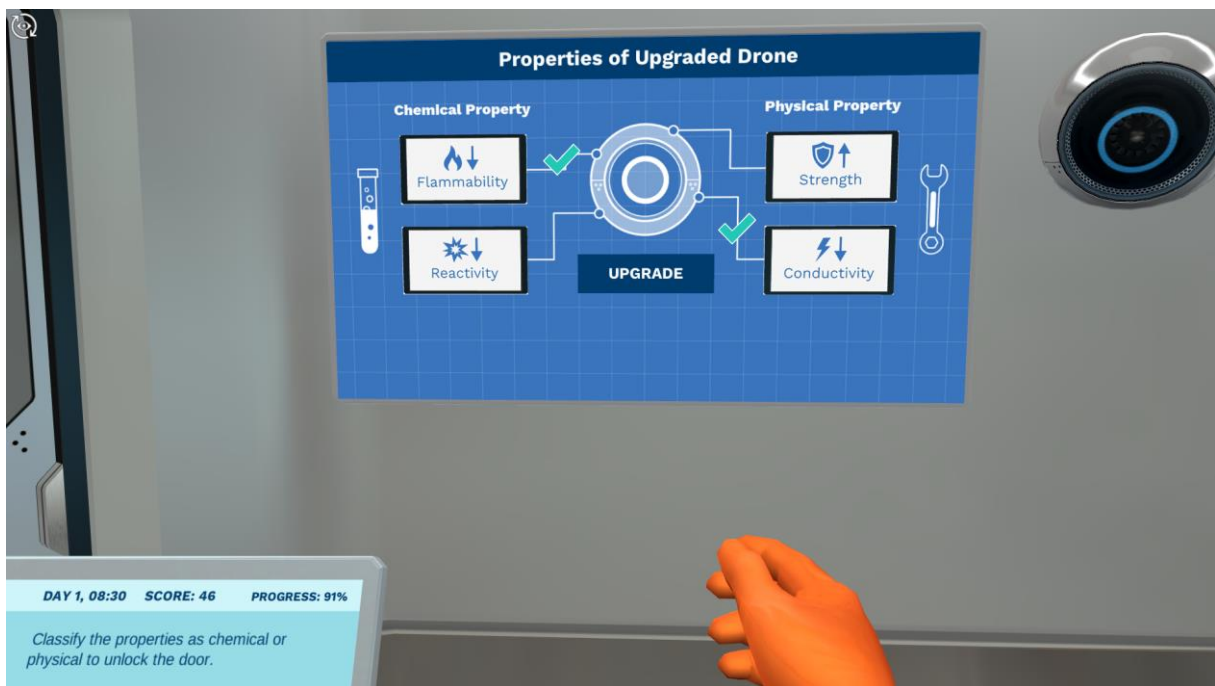
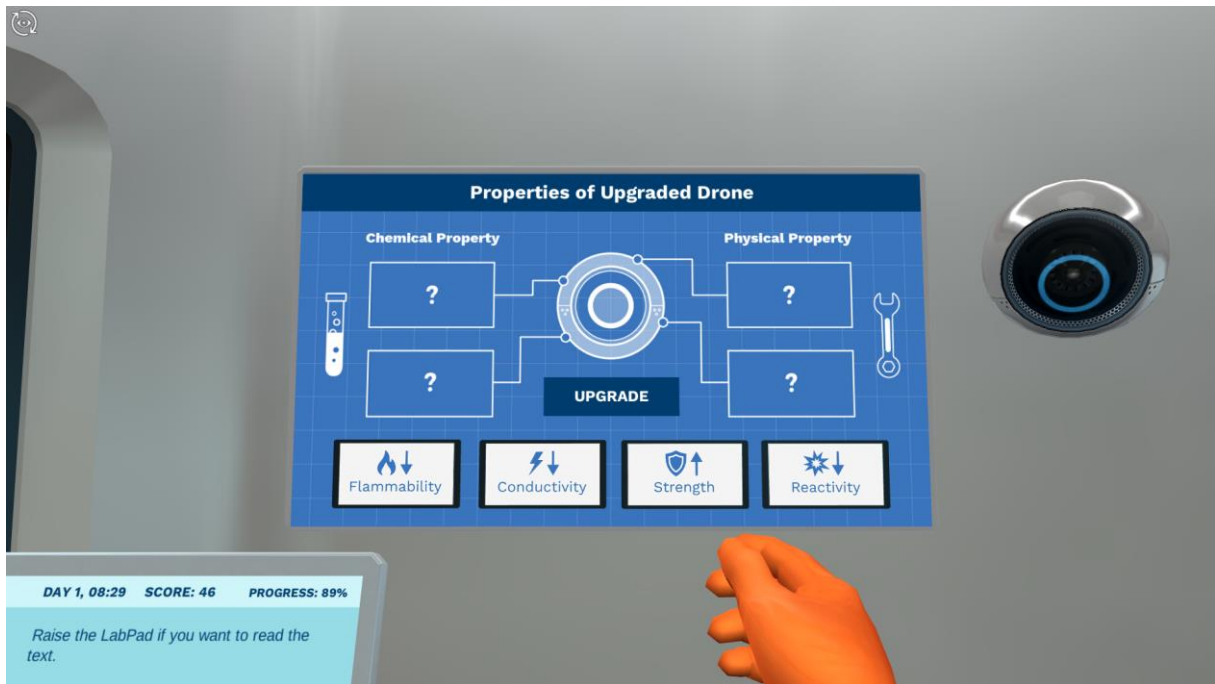
BACK



Neem elk voorwerp vast en doe het in de machine, druk op de oranje knop. Het voorwerp wordt geplet.



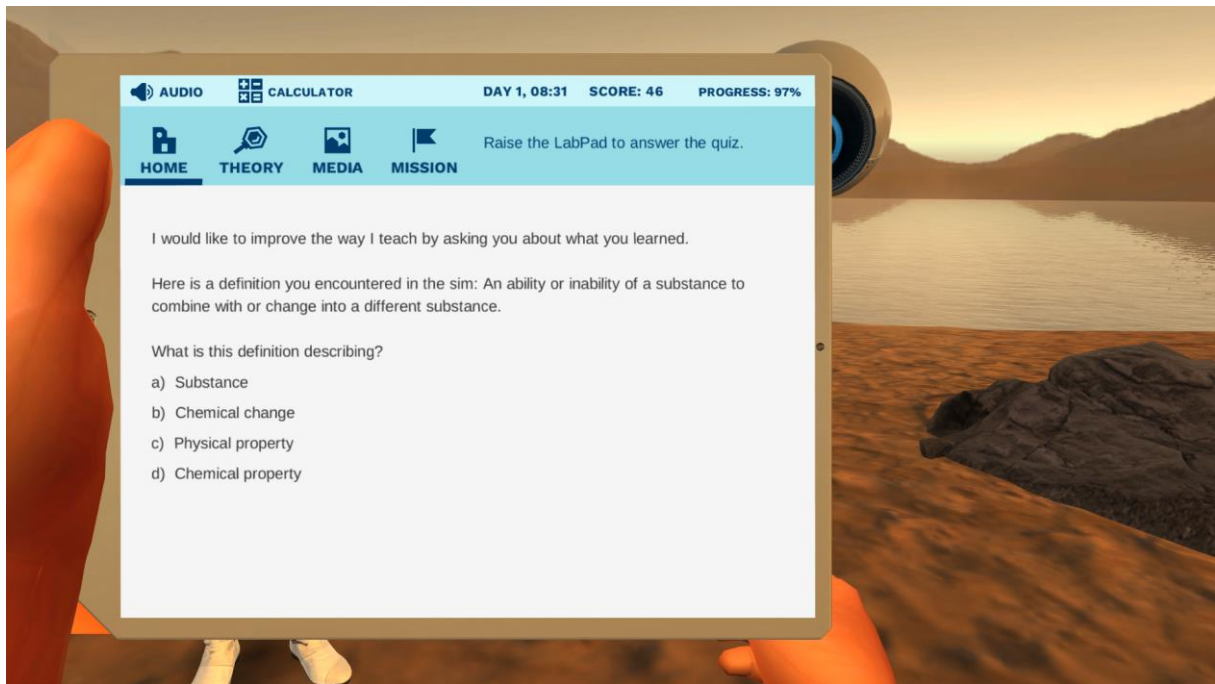
Brand het vuur.



Je wordt vervolgens geteleporteerd.

Tot slot krijg je een herhalingsvraag.





Antwoord c