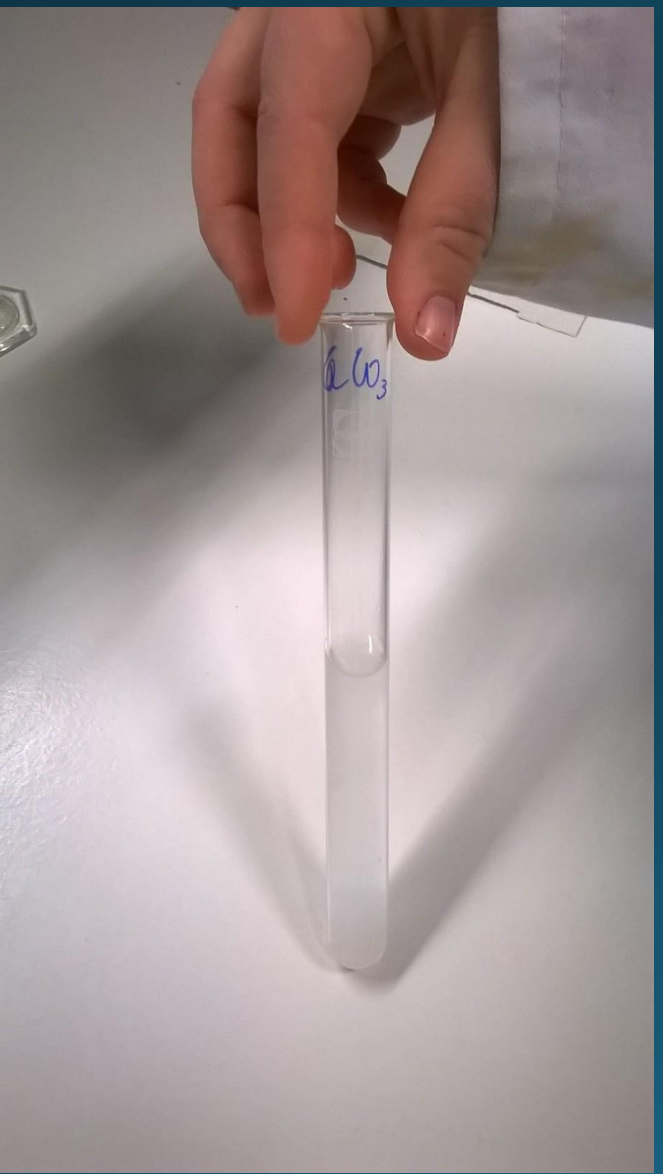
Ca^{2+} Neerslagen

Calciumdiwaterstofcarbonaat

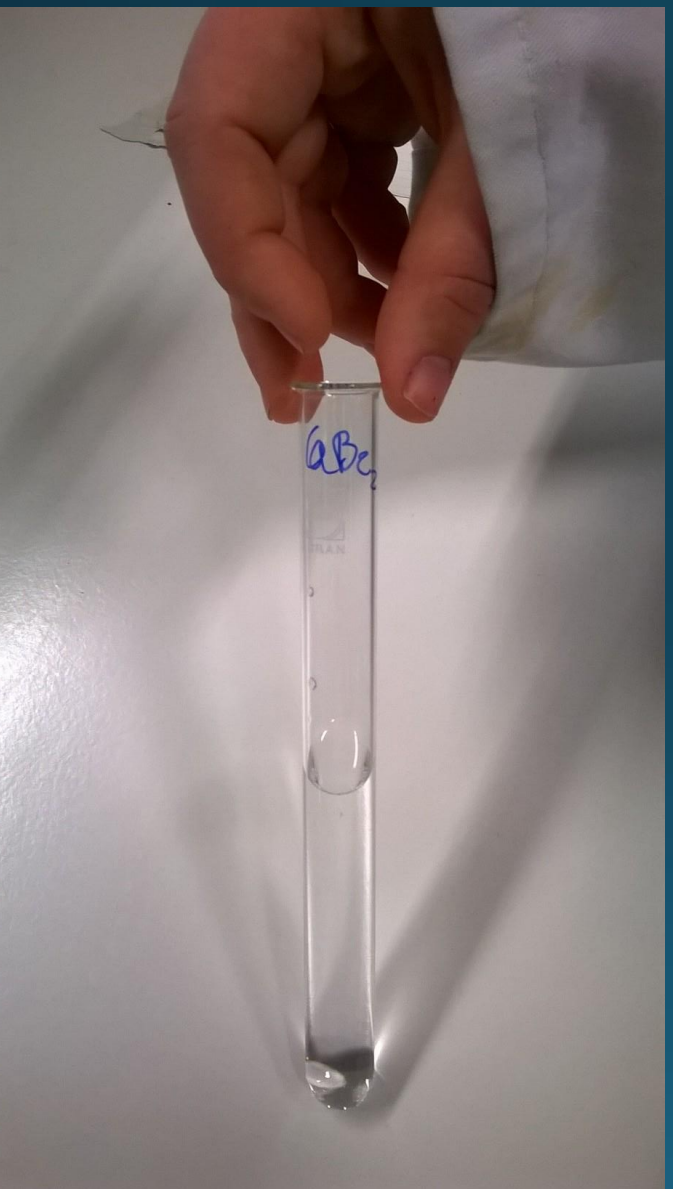
- $Ca^{2+} + 2HCO_3^- \rightarrow Ca(HCO_3)_2$
- vormt een neerslag

Calciumcarbonaat



- $Ca^{2+} + CO_3^{2-} \rightarrow CaCO_3$
- vormt een neerslag

Calciumdibromide

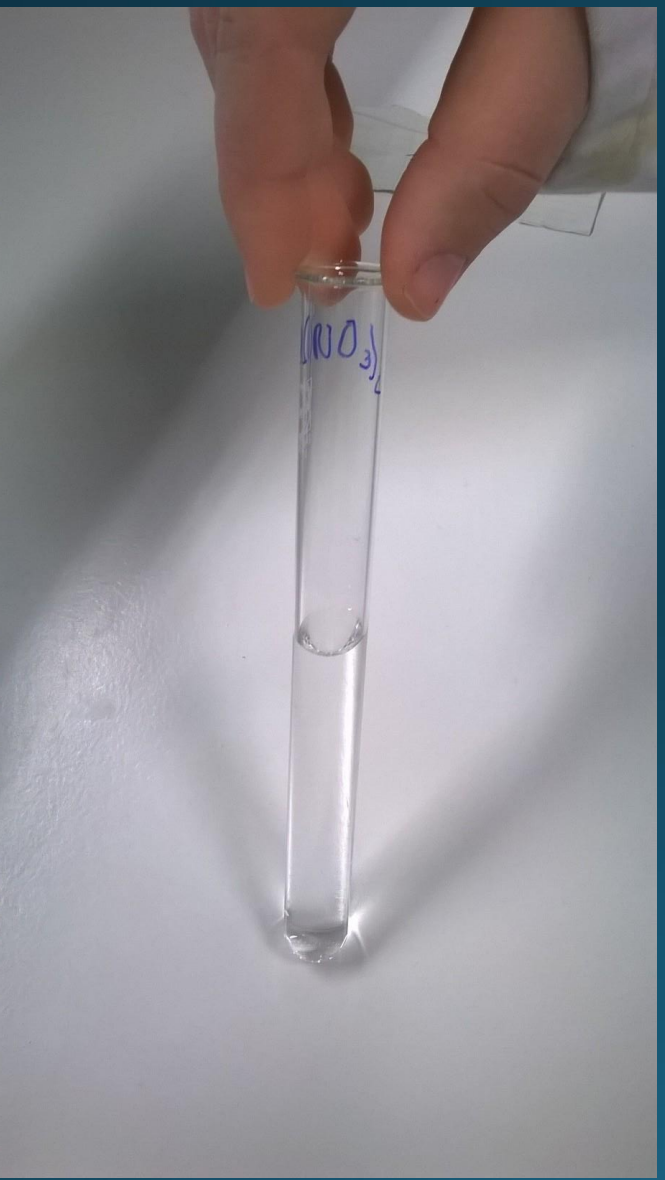


- $Ca^{2+} + 2Br^{-} \rightarrow CaBr_2$
- vormt geen neerslag

Calciumdichloride

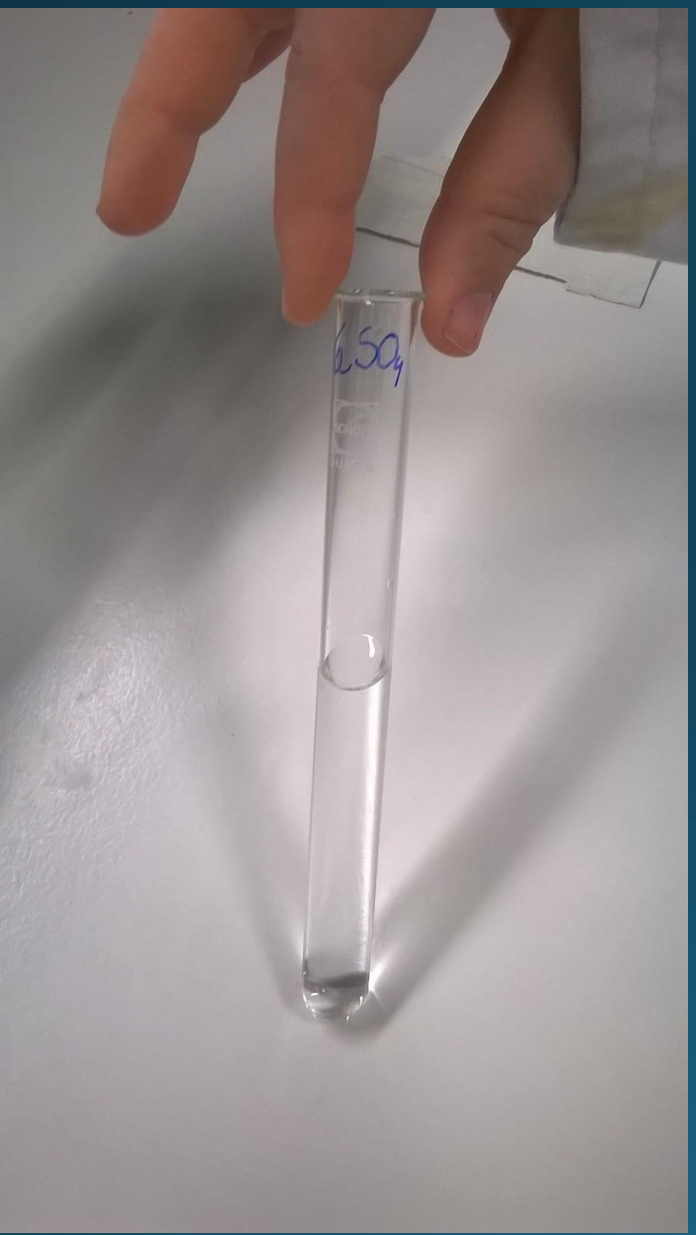
- $Ca^{2+} + 2Cl^{-} \rightarrow CaCl_2$
- vormt geen neerslag

Calciumdinitraat



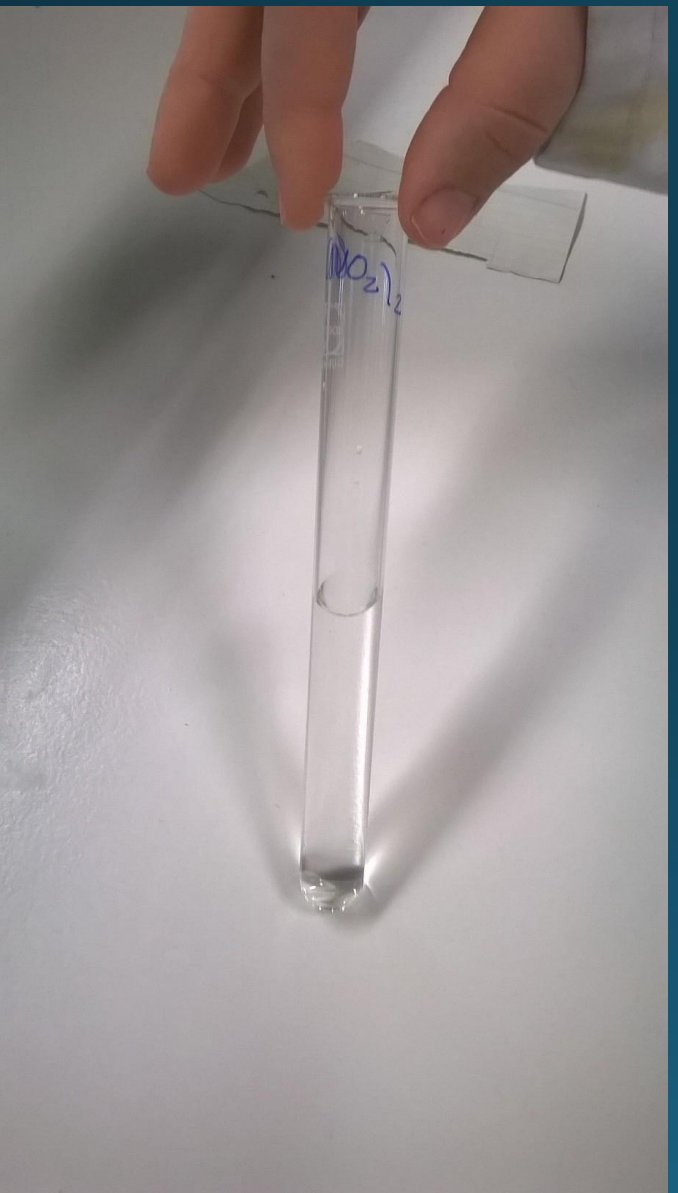
- $Ca^{2+} + 2NO_3^- \rightarrow Ca(NO_3)_2$
- vormt geen neerslag

Calciumsulfaat



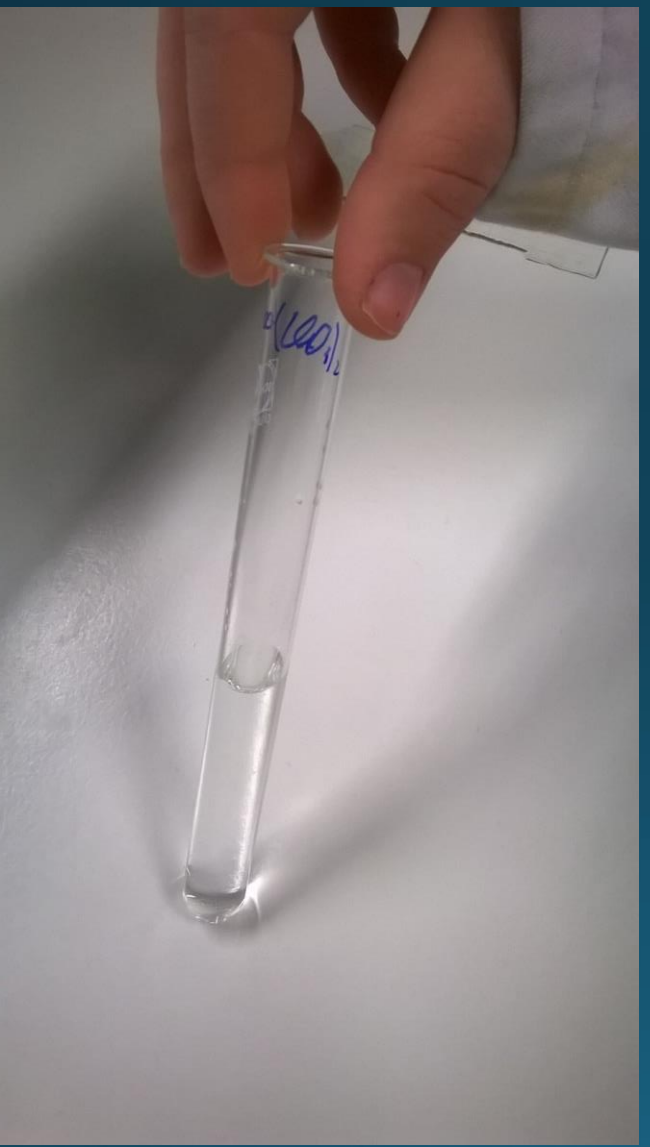
- $Ca^{2+} + SO_4^{2-} \rightarrow CaSO_4$
- vormt geen neerslag

Calciumdinitriet



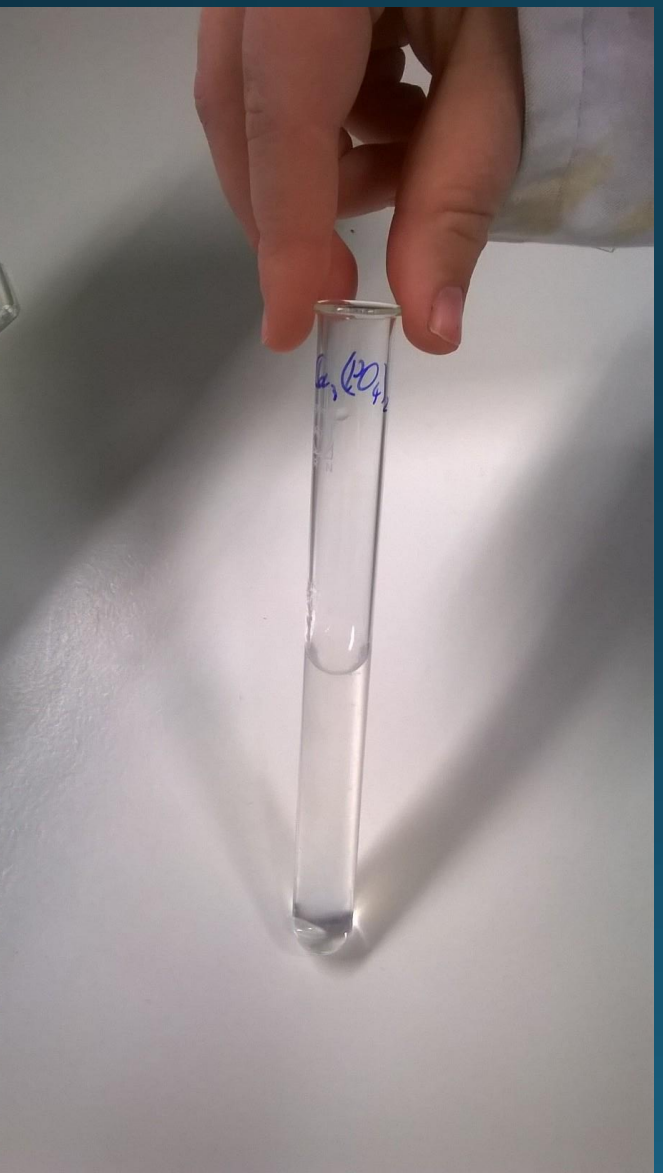
- $Ca^{2+} + 2NO_2^- \rightarrow Ca(NO_2)_2$
- vormt geen neerslag

Calciumdichloraat



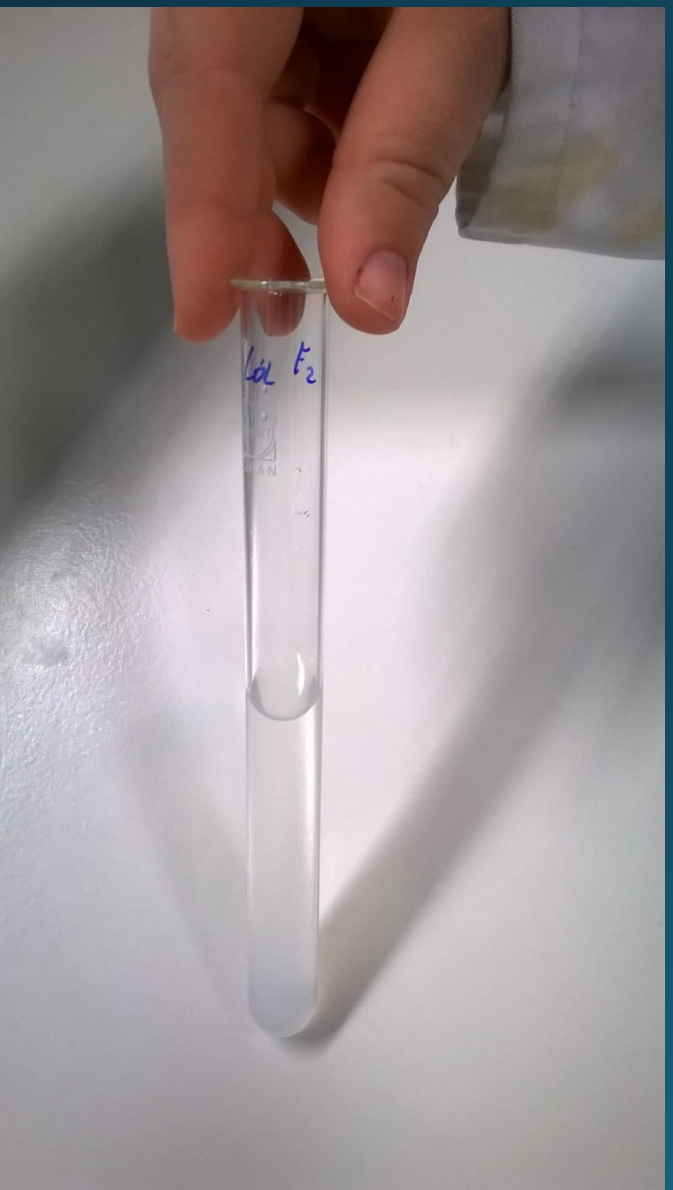
- $Ca^{2+} + 2ClO_3^- \rightarrow Ca(ClO_3)_2$
- vormt geen neerslag

Tricalciumdifosfaat



- $3Ca^{2+} + 2PO_4^{3-} \rightarrow Ca_3(PO_4)_2$
- vormt een lichte neerslag

Calciumdifluor



- $Ca^{2+} + F^{-} \rightarrow CaF_2$
- vormt een lichte neerslag

Calciumdijodaat

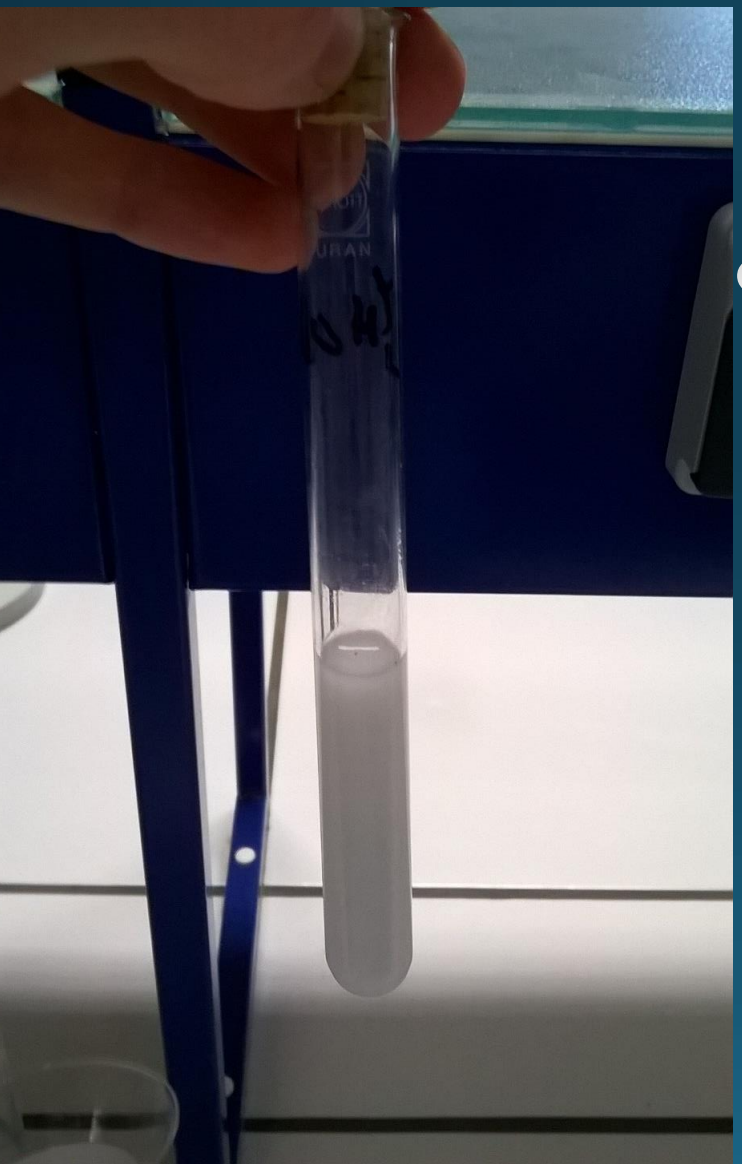
- $Ca^{2+} + IO_3^- \rightarrow Ca(IO_3)_2$
- vormt geen neerslag

Calciumsulfiet

- $Ca^{2+} + SO_3^{2-} \rightarrow CaSO_3$

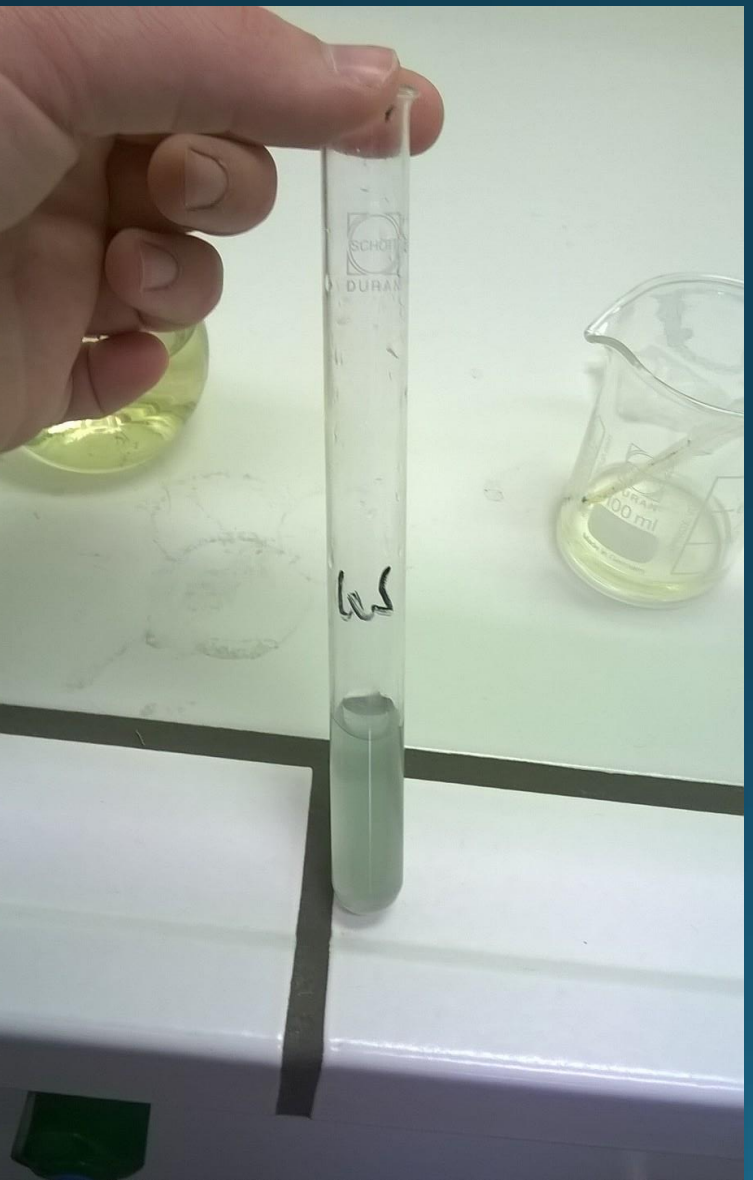
- vormt geen neerslag

Calciumdihydroxide



- $Ca^{2+} + OH^{-} \rightarrow Ca(OH)_2$
- vormt een neerslag

Calciumsulfide



- $Ca^{2+} + S^{2-} \rightarrow CaS$
- vormt een geelgroene neerslag