

# Neerslagreacties

## Zn<sup>2+</sup>

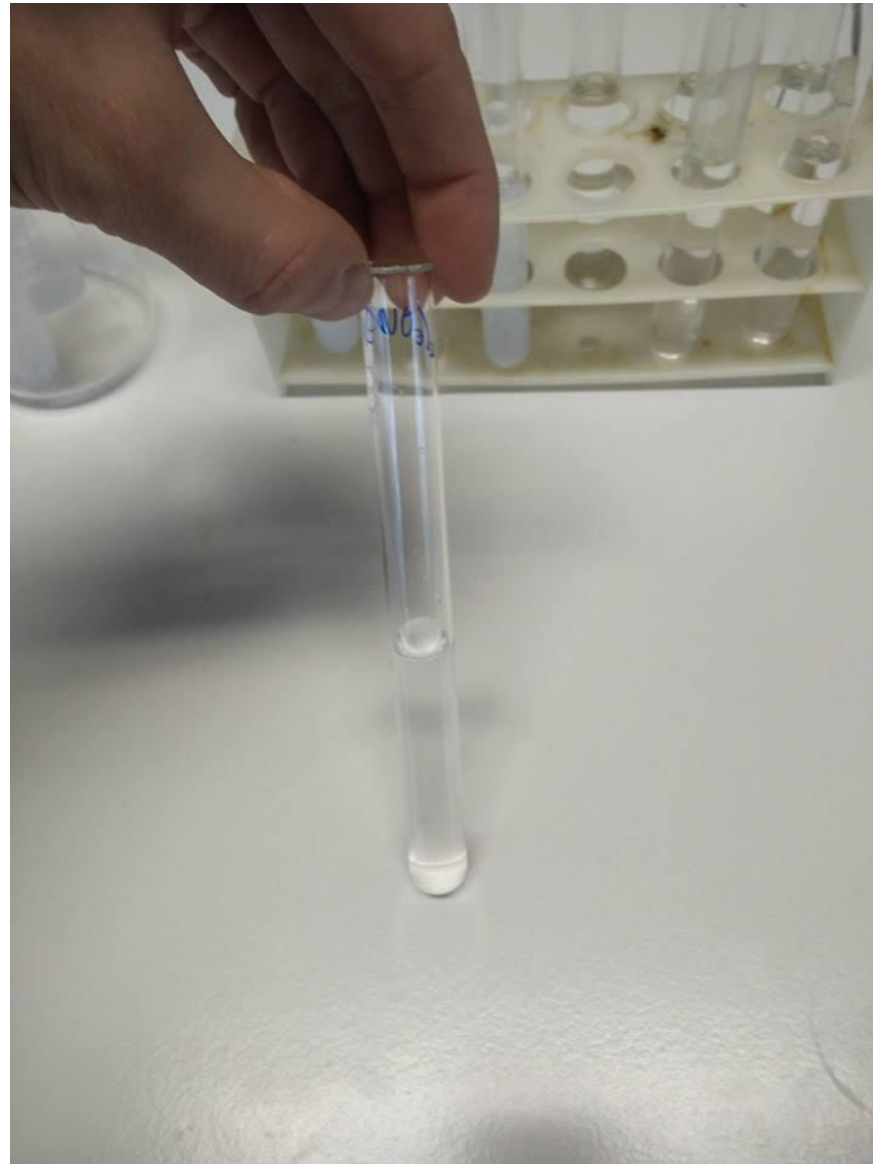
# Zinksulfaat

- ▶  $\text{ZnSO}_4$
- ▶  $\text{Zn}^{2+} + \text{SO}_4^{-} \rightarrow \text{ZnSO}_4$



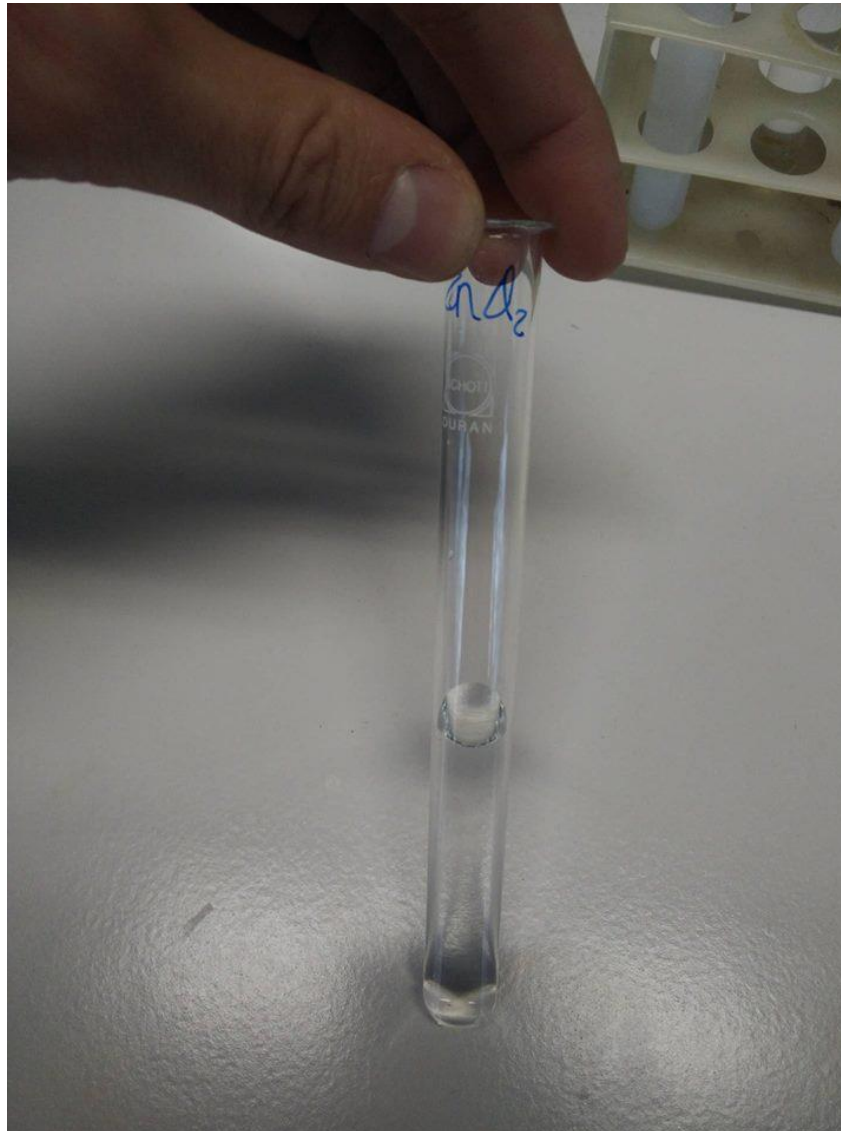
# Zinkdinitraat

- ▶  $\text{Zn}(\text{NO}_3)_2$
- ▶  $\text{Zn}^{2+} + 2\text{NO}_3^- \rightarrow \text{Zn}(\text{NO}_3)_2$



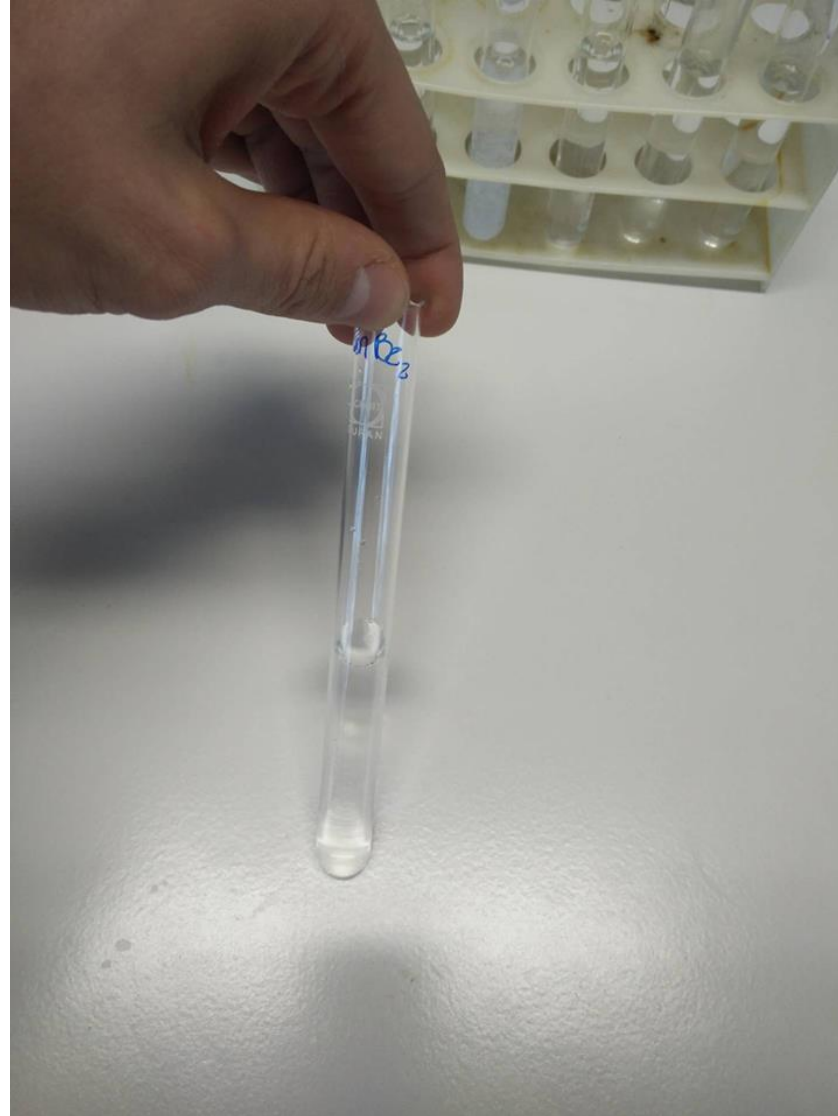
# Zinkdichloride

- ▶  $\text{ZnCl}_2$
- ▶  $\text{Zn}^{2+} + 2\text{Cl}^- \rightarrow \text{ZnCl}_2$



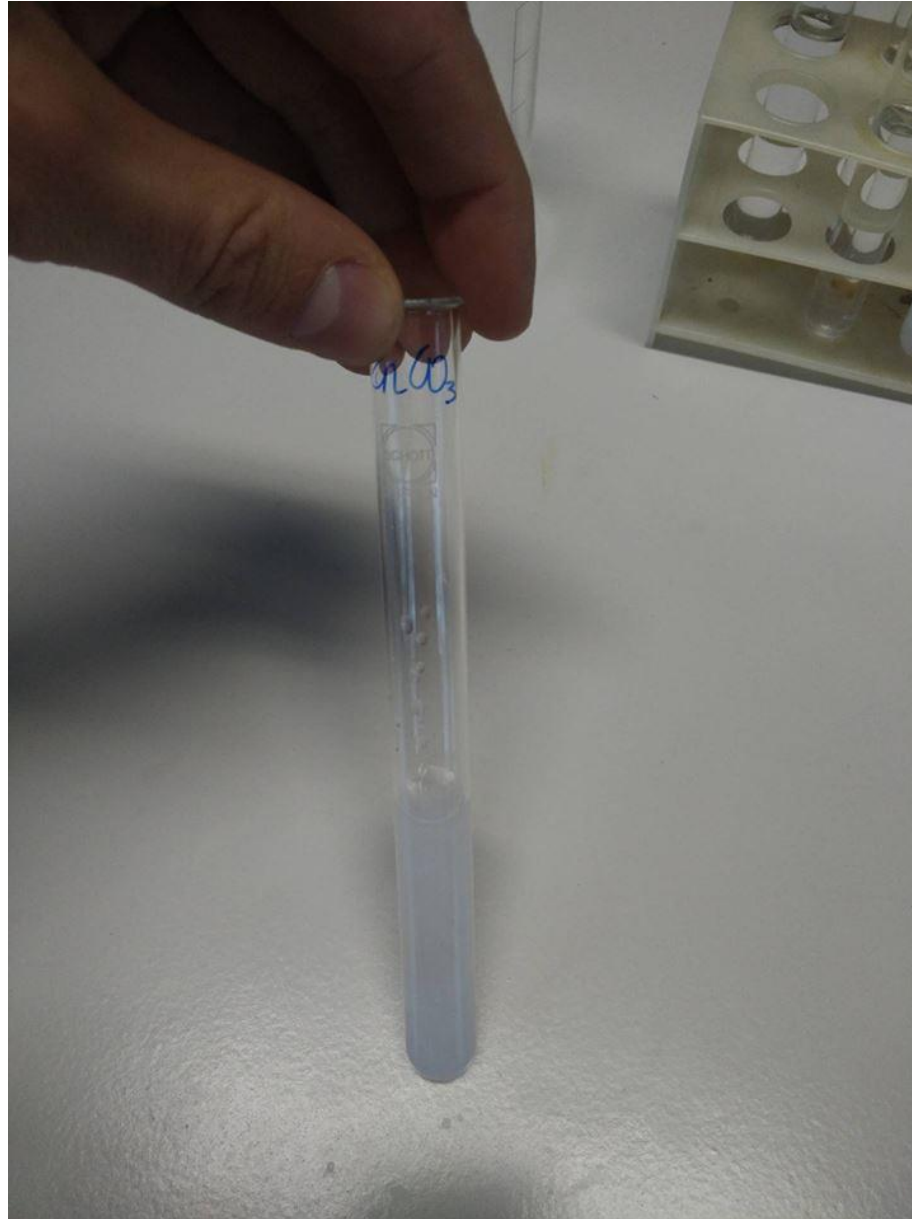
# Zinkdibromide

- ▶  $\text{ZnBr}_2$
- ▶  $\text{Zn}^{2+} + 2\text{Br}^- \rightarrow \text{ZnBr}_2$

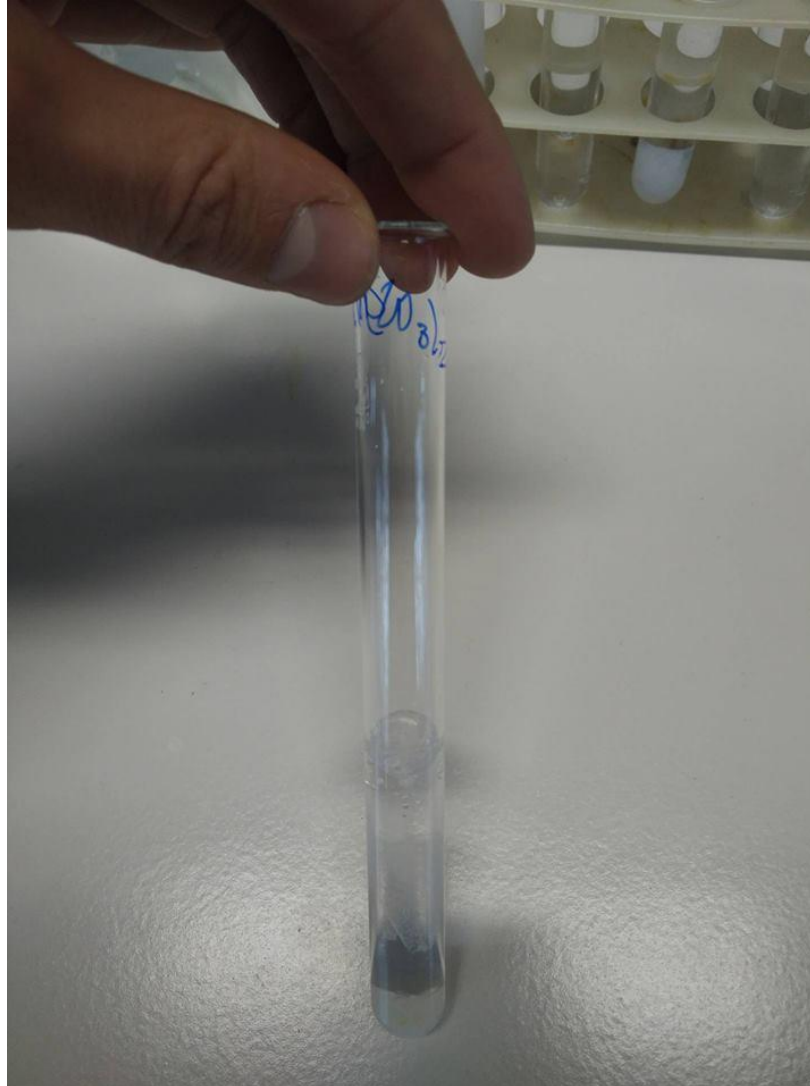
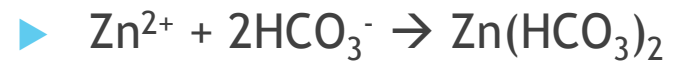
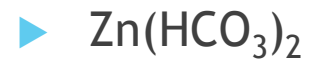


# Zinkcarbonaat

- ▶  $\text{ZnCO}_3$
- ▶  $\text{Zn}^{2+} + \text{CO}_3^{2-} \rightarrow \text{ZnCO}_3$



# Zinkdiwaterstofcarbonaat





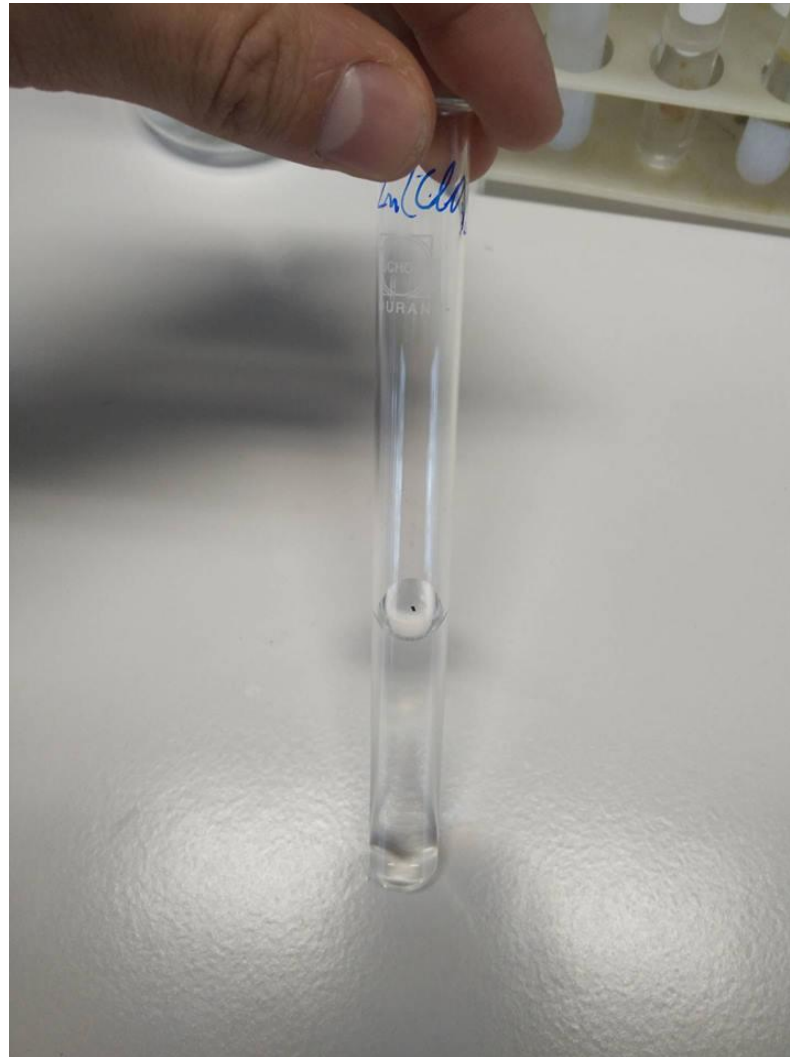
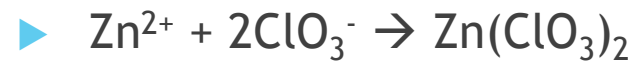
# Zinkdinitriet

- ▶  $\text{Zn}(\text{NO}_2)_2$
- ▶  $\text{Zn}^{2+} + 2\text{NO}_2^- \rightarrow \text{Zn}(\text{NO}_2)_2$

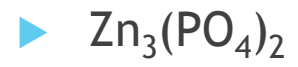




# Zinkdichloraat

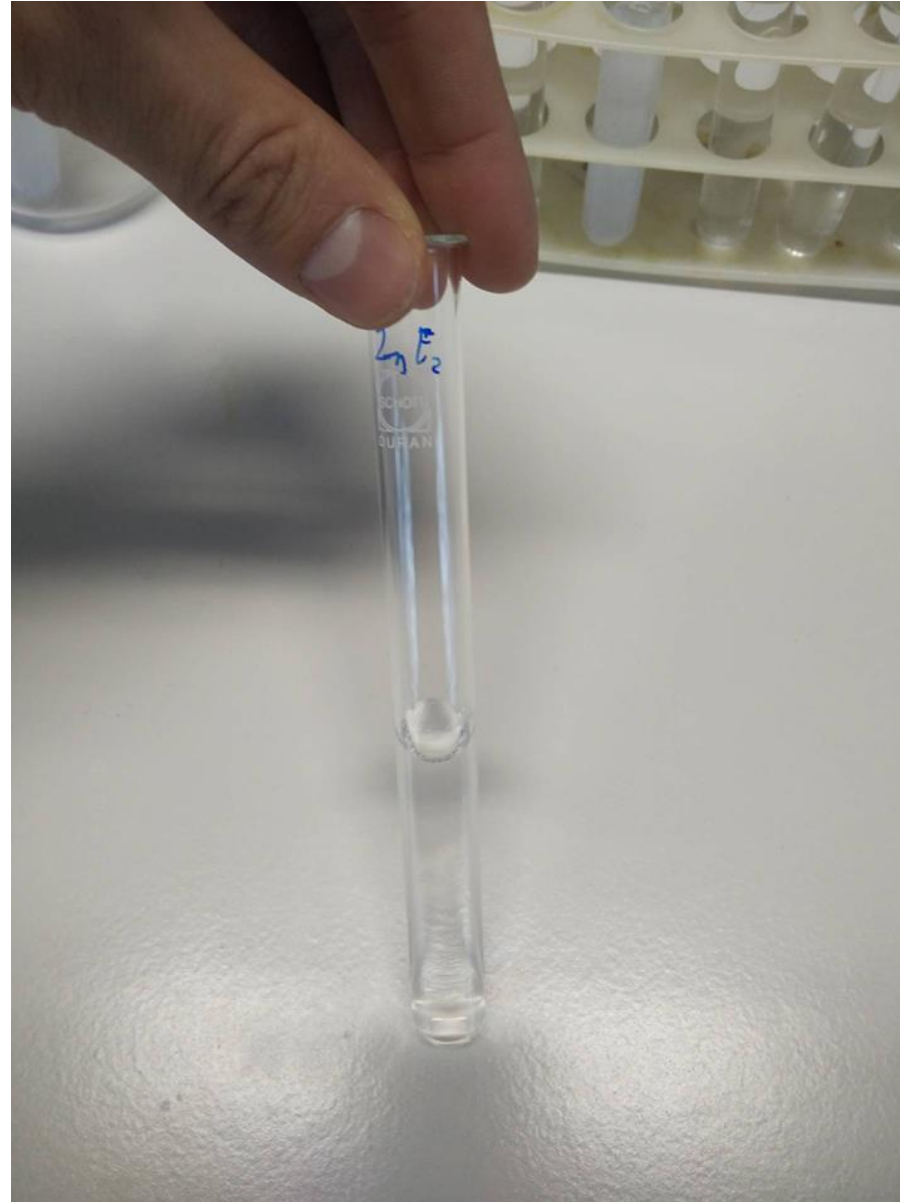


# Trizinkdifosfaat

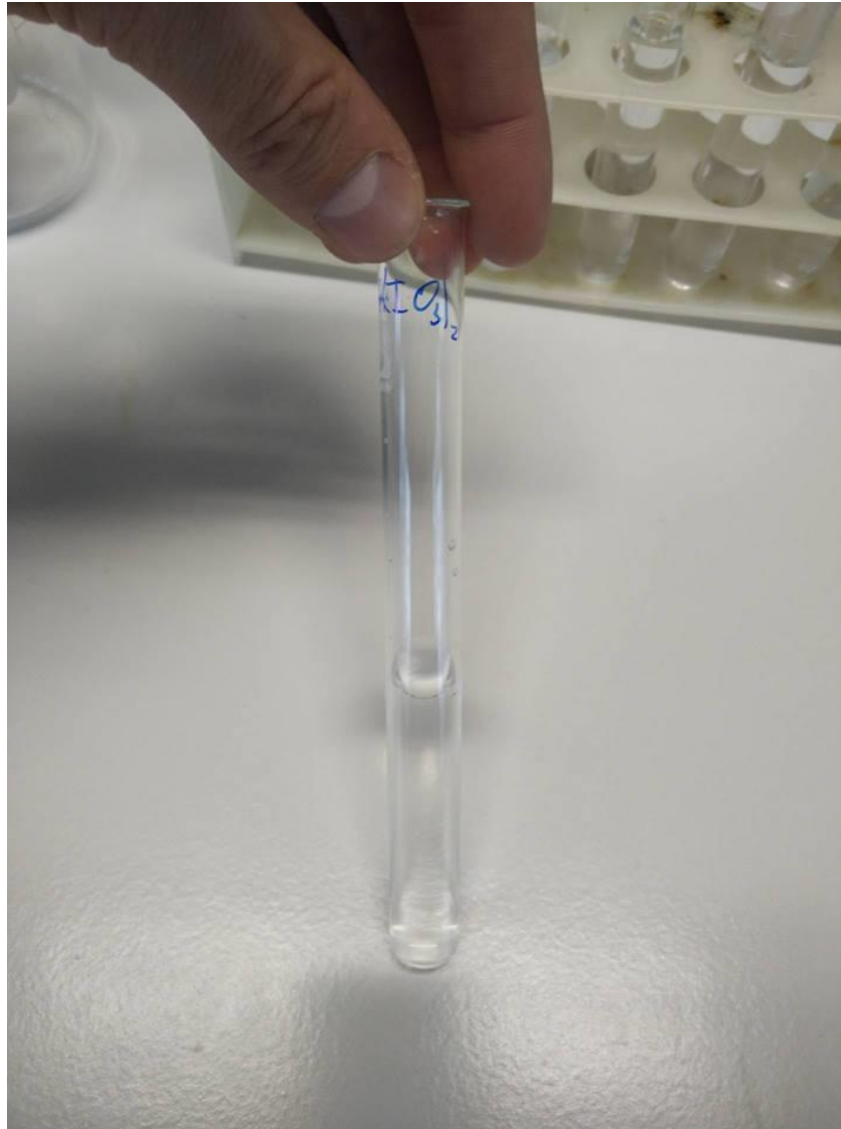
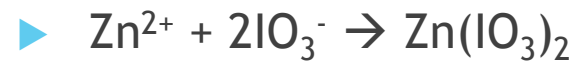
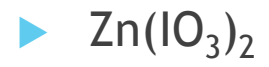


# Zinkdifluoride

- ▶  $\text{ZnF}_2$
- ▶  $\text{Zn}^{2+} + 2\text{F}^- \rightarrow \text{ZnF}_2$

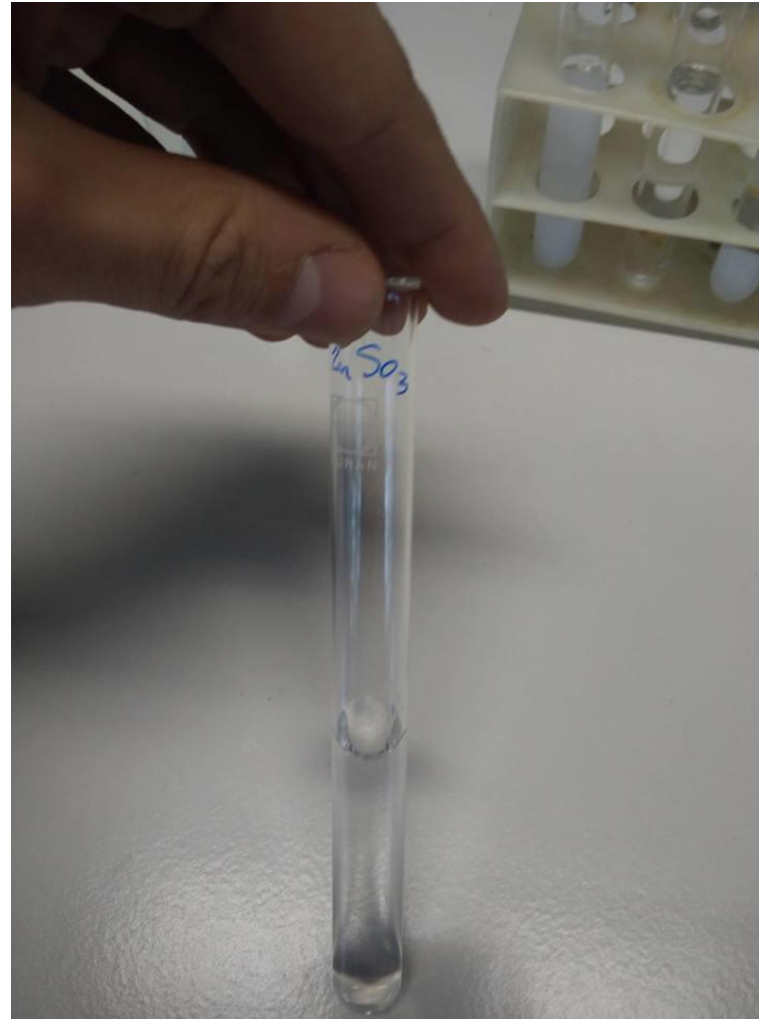


# Zinkdijodaat



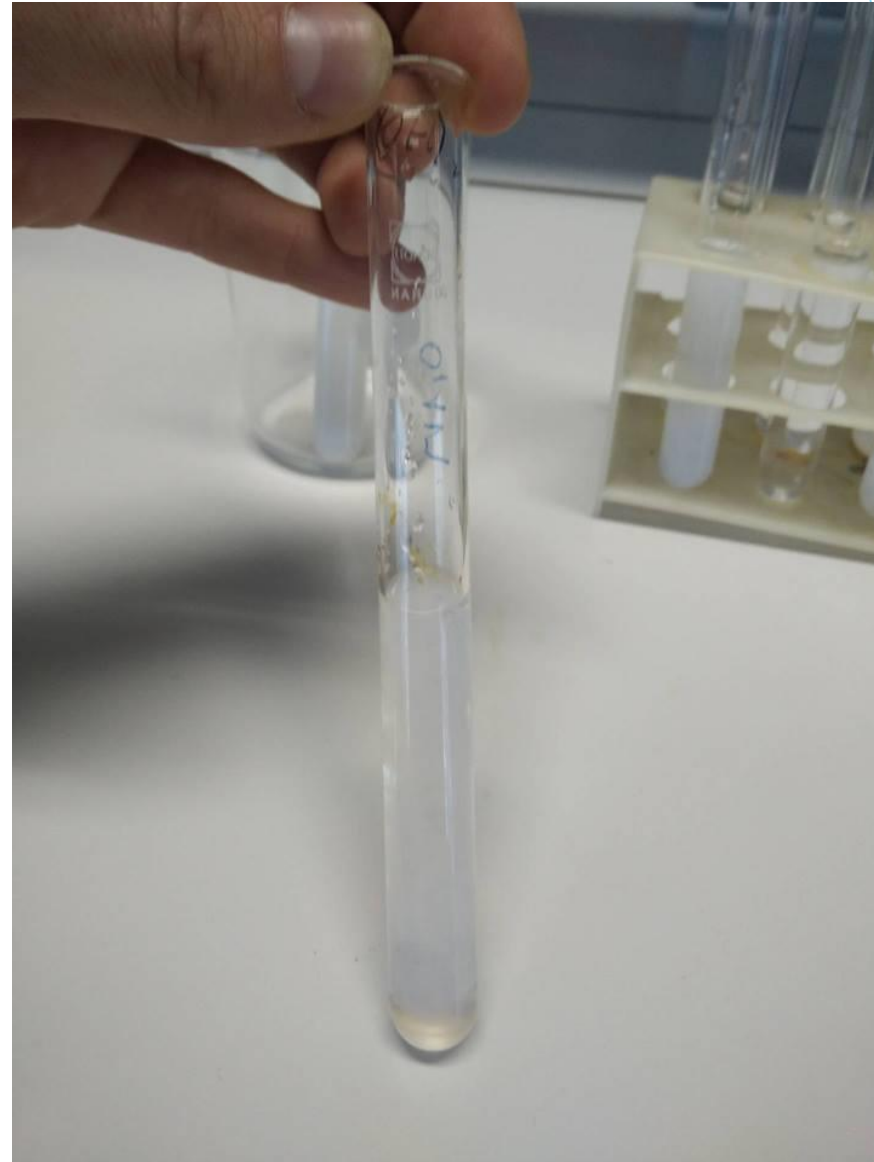
# Zinksulfid

- ▶  $\text{ZnSO}_3$
- ▶  $\text{Zn}^{2+} + \text{SO}_3^{2-} \rightarrow \text{ZnSO}_3$



# Zinkdihydroxide

- ▶  $\text{Zn(OH)}_2$
- ▶  $\text{Zn}^{2+} + \text{OH}^- \rightarrow \text{Zn(OH)}_2$



# Zinksulfide

- ▶ ZnS
- ▶  $\text{Zn}^{2+} + \text{S}^{2-} \rightarrow \text{ZnS}$

